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Industrial Sources Section
Source Management Branch
Robert S. Kerr Environmental Research Laboratory
Ada, Oklahoma 74820

ROBERT S. KERR ENVIRONMENTAL RESEARCH LABORATORY
OFFICE OF RESEARCH AND DEVELOPMENT
U.S. ENVIRONMENTAL PROTECTION AGENCY
ADA, OKLAHOMA 74820

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#### DISCLAIMER

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This report has been reviewed by the Robert S. Kerr Environmental Research Laboratory, U.S. Environmental Protection Agency, and approved for publication. Mention of trade names or commercial products does not constitute endorsement or recommendation for use.

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#### FOREWORD

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The Environmental Protection Agency was established to coordinate administration of the major Federal programs designed to protect the quality of our environment.

An important part of the Agency's effort involves the search for information about environmental problems, management techniques, and new technologies through which optimum use of the Nation's land and water resources can be assured and the threat pollution poses to the welfare of the American people can be minimized.

EPA's Office of Research and Development conducts this research through a nationwide network of research facilities.

As one of these facilities, the Robert S. Kerr Environmental Research Laboratory is responsible for the management of programs to: (a) investigate the nature, transport, fate and management of pollutants in ground water; (b) develop and demonstrate methods for treating wastewaters with soil and other natural systems; (c) develop and demonstrate pollution control technologies for animal production wastes; (e) develop and demonstrate technologies to prevent, control, or abate pollution from the petroleum refining and petrochemical industries, and (f) develop and demonstrate technologies to manage pollution resulting from combinations of industrial wastewaters or industrial/municipal wastewaters.

This report contributes to the knowledge essential if the EPA is to meet the requirements of environmental laws that it establish and enforce pollution control standards which are reasonable, cost effective, and provide adequate protection for the American public.

> W. C. Galegar Director

Robert S. Kerr Environmental Research Laboratory

#### ABSTRACT

This report is concerned with media disposition of specific priority pollutants. Composite samples were obtained from the influent, effluent, residuals, and air from 12 industrial biological treatment systems. These samples were extracted and analyzed by gas chromatography for organics constituents, by atomic absorption for metals, and by EPA methodology for phenolics, cyanide, and mercury.

Participating industries include: (1) organics and plastics, (2) pharmaceuticals, (3) pesticides, (4) rubber, (5) wood preservative, and (6) petroleum refining. Each of the 12 cooperating companies reviewed and commented on the draft report for description of its biological treatment system, accuracy of the study conditions as well as comments on the completed analytical data. Data completed after issuance of the draft report have since been submitted to the company officials for additional comments.

This report covers a period from May 1978 to February 1979, and work was completed as of March 1979.

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### LIST OF ABBREVIATIONS AND SYMBOLS

# **ABBREVIATIONS** -- cubic feet per hour\* cfh -- Effluent Guidelines Division EGD -- Environmental Protection Agency EPA --- gallons per minute\* gpm -- million gallons\* mg mgd -- million gallons per day\* -- milliliter ml -- milligrams per liter mg/1-- powdered activated carbon PAC -- Robert S. Kerr Environmental Research Laboratory **RSKERL** VOA -- volatile organics analysis **SYMBOLS** -- micron -- micrograms per liter μg/l -- microgram per kilogram μg/kg

\*See Table 37, Metric Conversion Table, on page 89

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# ACKNOWLEDGEMENTS

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To successfully complete any applied research study, the research team needs the full cooperation of all interested parties. Our acknowledgement of appreciation is extended to EPA's Office of Research and Development (OEMI and OALWU) and Office of Water and Hazardous Materials (EGD) for participation in planning and administrative tasks associated with the study. A special acknowledgement is due the industrial plants—their executives, supervisors, and operators—for the excellent cooperation extended to the RSKERL teams during this study.

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This document is a preliminary draft. It has not been formally released by the U.S. Environmental Protection Agency and should not at this stage be construed to represent Agency policy. It is being circulated for comments on its technical merit and policy implications.

# SECTION 1

#### INTRODUCTION

In June 1978, the Robert S. Kerr Environmental Research Laboratory (RSKERL) began an applied research study at the request of EPA's Effluent Guidelines Division (EGD). The purpose of the study was to determine what happens to specific priority pollutants as they pass through a biological treatment system. A mass balance study on a biological treatment system would consume considerable time, manpower, and resources. It was therefore decided that an indication of the fate of specific priority pollutants would suffice in lieu of a mass balance study. This "Indicatory Fate Study," planned and conducted by RSKERL, represents a screening study to view the removal of specific priority pollutants via air, water, or residuals routes.

Specific organic pollutants were analyzed by gas chromatography using methodology supplied by EGD. Quality control for the specific organic priority pollutants was provided by gas chromatograph/mass spectrometer analysis on one sample obtained from each plant site visited. Total metals, phenol, and cyanide analyses were conducted in accordance with EPA's April, 1977, Protocol.

Twelve industrial participants representing six industrial categories were selected by Effluent Guidelines: (1) organic chemicals, (2) petroleum refining, (3) pharmaceuticals, (4) wood preserving, (5) pesticides, and (6) rubber. Based on analytical data generated during previous screening studies, EGD requested sampling for specific priority pollutants from each participant.

Responsible company officials for each plant selected were notified by a letter from the Director, RSKERL, explaining the purpose of the proposed study and requesting their participation. Each of the plants visited cooperated fully in completing the field sampling portion of the study.

The data in this report provide only an indication of the route of removal of specific priority pollutants and are not intended to represent a mass balance across a biological treatment system.

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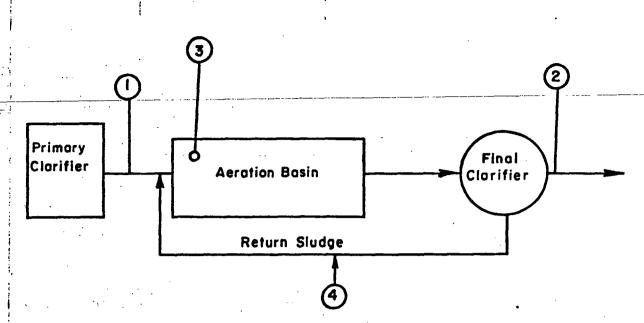
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Figure 1. Activated Sludge Sampling Points

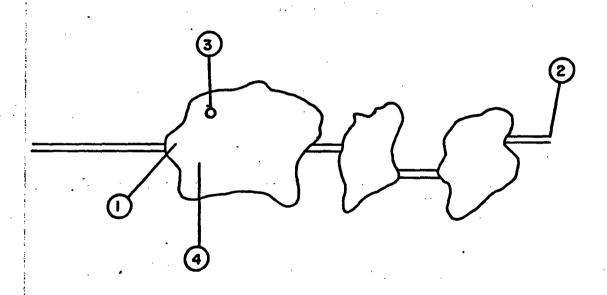


Figure 2. Aerated Lagoon Sampling Points

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### SECTION 2

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### SAMPLING-PROGRAM

The sampling program was designed to examine three possible removal routes for each type of treatment system: (1) air, (2) water, and (3) residuals.

### SAMPLE LOCATIONS

Two basic types of biological treatment systems were studied: (1) activated sludge and (2) aerated lagoons.

Sampling points established to examine removal routes from activated sludge treatment systems were: (Figure 1)

•	Sample Location	Sample Type	<u>Media</u>
1.	Primary Effluent	72-hr Composite, VOA Grab	Water
2.	Final Effluent	72-hr Composite VOA Grab	tt .
3.	Aeration Basin	Stripper Sampler Composite	Air
4.	Return Sludge	72-hr Composite	Residuals

Sampling points established for aerated lagoon treatment systems were: (Figure 2)

4.	Bottom Deposits	72-hr Composite	Residuals
3.	Near first Lagoon Inlet	Stripper Sampler Composite	Air
2.	Outlet from Last Lagoon	72-hr Composite VOA Grab	tt
1.	Inlet to First Lagoon	72-hr Composite VOA Grab	Water
	Sample Location	Sample Type	Media

 $\{x_i, x_i\}^{t}$ 

# Air (Figure 3)

Commensurate with the beginning of the sampling program, an air sampler was installed in the aeration basin (mixed liquor) of the activated sludge treatment system or in the first lagoon of the aerated lagoon treatment system. The air sampler was designed at RSKERL for this study and is a combination stripper/adsorber. A 20-gallon/minute submersible pump continuously supplied a fresh sample of water to the stripper, while a carbon-filtered air supply was used to sparge volatile organic compounds from the water sample. The sparged air was fed to a slurry pot containing XAD2 resin for adsorption of the volatile organic compounds. The sparged air was also fed intermittently to a gas chromatograph column packed with Tenax for adsorption of light volatile hydrocarbons.

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### Water

Water samples were collected at the inlet and effluent points of the biological treatment system. A 3-day composite sample was prepared at each sample point by collecting 24 equal aliquots at specified times during the 72-hour sampling period.

In addition, VOA grab samples were collected at the influent and effluent sampling points. These samples were collected in previously prepared 30-ml sample vials. Blank VOA vials containing Super Q water were uncapped prior to, and resealed following, collection of VOA samples at each location. The VOA samples were used to supplement samples collected using the airstripper sampler.

## Residuals

At plants employing an activated sludge system, the return sludge was sampled in the same mode as the water samples. At those plants where an aerated lagoon was being studied, a single dredge sample was obtained near the inlet to the first lagoon.

#### SAMPLE HANDLING

Samples were preserved in accordance with the April, 1977, Protocol. Water and residual samples were kept iced for the duration of the sampling program. In addition, metal samples were preserved with redistilled nitric acid; phenol samples were

- (R) Registered trademark Rohm & Haas (XAD-2)
- (R) Registered trademark ENKA N.V., Holland (Tenax)

PACE MANAGES

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Figure 3 - AIR-STRIPPER SAMPLER

Foam Knock-out

XAD-2 Resin

Slurry

Trap

Sub. Pump for mixed-liquor recirculation

Cleaned Compressed

Rotameter (2)

Filter

**Activated** 

Comp.Air

Carbon

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preserved with phosphoric acid; and cyanide samples were preserved with sodium hydroxide. No additional preservative was used for the organics samples.

After the sampling study was complete, sample chests, filled with ice, were transported to RSKERL, where they were transferred to a 4°C walk-in constant-temperature box. The samples remained in this environment until they were ready to be extracted and/or analyzed.

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# SECTION 3

### PLANT STUDIES

### ORGANICS AND PLASTICS INDUSTRY

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### Plant 1

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Wastewater Treatment System --

A flow diagram of Plant 1's wastewater treatment system is shown in Figure 4. Wastewater generated by the manufacturing of organic chemicals and plastics is treated by a series of processes that generally consist of (1) neutralization, (2) denitrification, (3) aeration, and (4) nitrification. Sludge produced by the wastewater treatment process is digested aerobically, filtered in a dual-cell gravity unit, and landfilled. An approximate material balance for the treatment system is shown in Table 1.

Raw wastewater from the manufacturing process is first neutralized in a limestone digestor. Here the pH of the raw wastewater is increased from about 1.5 to about 4.8. From the neutralization pit, the wastewater flows to a 1-million-gallon denitrification pond. Here the wastewater is mixed with a portion of the return sludge. A single 75-horsepower (hp) aerator is used to mix the contents of the denitrification pond. Some aeration does occur; however, the amount of oxygen transferred is low enough that the pond remains essentially anaerobic.

From the denitrification section, the wastewater flows to a 1.1-million-gallon aeration basin, where another portion of the return sludge is added to the wastes, and is aerated using three mechanical aerators. One aerator is a variable speed, 150-hp aerator and is fixed mounted. The other two aerators are floating, high-speed, 50-hp aerators. From the aeration section, the wastewater flows to a 1.1-million-gallon nitrification pond, where the final portion of return sludge is added to the wastewater, and is aerated with three mechanical aerators. One aerator is a fixed speed, 150-hp aerator and is fixed mounted. The other two aerators are floating, high-speed, 50-hp aerators.

The wastewater is then clarified, and the effluent is discharged into a local river. The total retention time of the treatment system is 1.48 days at 1,500 gallons per minute (gpm), and it has a total volume of 3.2 million gallons.

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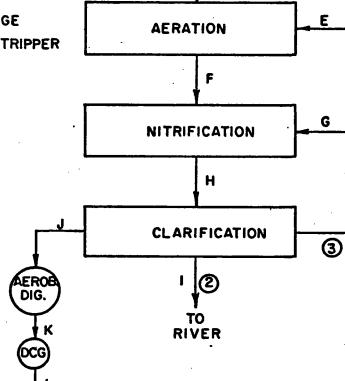


Figure 4 - WASTEWATER TREATMENT SYSTEM - PLANT I

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TABLE :	<ol> <li>BIOLOGICA</li> </ol>	L TREATMENT PLAN	T MATERIAL BALANCE-	PLANT 1

Development states

	Stream		TOC	IC #/hr.	NH <sub>3</sub> -N #/hr.	NO <sub>3</sub> -N #/hr.		рН	0 <sub>2</sub> #/hr.
Total Influent	· <b>A</b>	952,000	490	9	84	350	180	1.5	-
Digestor Effluent	<b>B</b>	952 <b>,</b> 700	490	85	84	350	180	4.8	_
Clarifier Recycle	C	605,000	24	163	9	36	9,070	8.0	-
Denitrification	D 1,	545,700	190	480	93	42	9,250	8.2	-
Clairfier Recycle	E	305,000	12	82	5	18	4;570	8.0	· -
Aeration	F 1,	850,700	74	594	65	77	7,570	8.1	230
Clarifier Recycle	G	460,000	18	124	7	28	6,900	8.0	-
Nitrification	н 2,	<b>310,</b> 700	92	62	35	139	20,914	8.0	150
Effluent	I	937,700	37	250	14	56	60	8.0	-
Wasted Sludge	J	15,000	-	-	-		211	8.0	-
Aerobic Digestor	K	15,000	· _	-	-	-	211	7.5	-
Land-Fill	L	3,000	-	-	-		210	7.5	_

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Sample Collection --

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Based on previous screening conducted at Plant 1 under the direction of EPA's Effluent Guidelines Division, a list of priority pollutants was compiled for investigation in this study (Table 4). This list represents the priority pollutants which have been identified in Plant 1's influent to the treatment system. From this list, it was determined that six 1-gallon samples of composite were required for specific organic compound analysis.

The sampling point locations are shown in Figure 4. The influent sample was taken immediately following the limestone digestion pit. The effluent sample was taken out of the clarifier overflow channel. The return sludge sample was taken off the discharge side of the recycle pump. The air-stripper sampler was placed just off the walkway in the denitrification pond.

Samples of the influent, effluent, and return sludge were composited. Every 3 hours, beginning at 9 pm on July 10, 1978, six 130-milliliter (ml) aliquots were taken at each sample point for specific organic analysis. In addition, three 40-ml aliquots were taken for cyanide, phenolics, and metals. Each metals, cyanide, and phenolics sample was "preserved" at 10:30 am on July 11, by adding 5 ml of nitric acid to the metals samples, 1 ml of sodium hydroxide to the cyanide, and 2 ml of phosphoric acid to the phenolics samples. At all times each of the samples taken was kept on ice. At 5 pm on July 13, 1978, VOA grab samples were taken for the influent and effluent. A detailed sampling schedule is presented in Table 2.

The air-stripper sampler was placed into operation at 9:30 pm on July 10, 1978, and operations were concluded at 6 pm on July 13, 1978, for a total operating time of 68 1/2 hours. Air charged to the stripper averaged 60 cubic feet per hour (cfh), and the quantity of air to the XAD-2 scrubber was 30 cfh. Stripped air was collected on the Tenax columns for 10 minutes each day. At this plant two Tenax columns were used. On the first day one column was used; for the second and third days another column was used.

Table 3 contains flow data for this plant's treatment system on the days sampled.

Analytical Results--

Priority pollutants for which samples were collected and analyzed are presented in Table 4. All extractions and analyses of samples were conducted at RSKERL.

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TABLE 2. SAMPLING SCHEDULE (PLANT 1)

Minis Chas Car

Date	Time	Samples taken	Remarks
7/10/78	9:00 pm 12:00 Midnight	Composites*	
7/11/78	3:00 am 6:00 am 9:00 am		Added 5 ml HNO3, 2 ml
	10:30 am	11	NaOH, 2 ml H <sub>3</sub> PO <sub>4</sub> to metals, cyanide, and phenolics composites
11 11 11 11	12:00 Noon 3:00 pm 6:00 pm 9:00 pm 12:00 Midnight	Composites	
7/12/78	3:00 am 6:00 am 9:00 am 12:00 Noon 3:00 pm 6:00 pm 9:00 pm	11 11 11 11 11 11 11	•
7/13/78	3:00 am 6:00 am 9:00 am 12:00 Noon 3:00 pm 5:00 pm 6:00 pm	VOA's Composites	

<sup>\*</sup>Composites consist of six 130-ml aliquots for organics, three 40-ml aliquots for metals, cyanide, and phenols.

TABLE 3. DAILY FLOW DATA (PLANT 1)

Date	Time	Flow (gpm)	
7/10/78	Daily average	1,623	
7/11/78	11	1,404	•
7/12/78	##	1,698	
7/13/78	8 am	1,590	
7/13/78	12 Noon	2,340	
7/13/78	4 pm	2,280	

TATE WAS

•	Priority Pollutant	Influent (µg/1)	Return Sludge (µg/1)	Effluent (µg/1)	
1	CLASSICAL				
	TOTAL CYANIDES (mg/1)*	.62	.16	<.08	
	TOTAL PHENOL	21	<b>&lt;</b> 50	<20	
•	TOTAL METALS				
•	Arsenic	12	200	<10	
	Selenium	<10	<10	<10	
	Cadmium	3	48	1	
	Beryllium	<3	· 11	· <b>&lt;3</b>	
	Copper	160	4,000	17	
	Antimony	<10	<10	<10	
·	Chromium	650	18,000	50	
	Nickel	· · · 81	3,900	65	
	Zinc	770	15,000	89	
	Silver	<10	13	<10	
	Thallium	<10	<10	<10	
	Lead	10	530	<10	
	Mercury	<1.0	<.5.0	05	
	ORGANICS (GAS CHROMATOGRAPHY)				
	PURGEABLES				
	1,4-Dichlorobenzene	<10	N.A.	<10	
	Benzene	405	N.A.	<40	
•	1,3-Dichlorobenzene	<10	N.A.	<10	
	Chloroform	<10	N.A.	<10	
	1,2-Dichloropropane	<10	N.A.	<10	
	Methylene chloride	<10	N.A.	<10	
	Ethylbenzene	<10	N.A.	<10	
	Trans-dichloroethylene	<10	N.A.	<10	
	DINITROTOLUENE				
•	Nitrobenzene	N.P.	N.P.	N.P	
	HALOETHER				
	2-Chloroethyl vinyl ether				
	POLYNUCLEAR AROMATICS				
-	Phenanthrene/Anthracene	<50	<50	<50	
	Fluorene	<64	<64	<64	
	Naphthalene	<22	<22	<22	
30.F1					
A GLUS	(Continued)				

12 PAGE AL CON

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(Continued) TABLE 4. Influent Return Sludge Effluent Priority Pollutant  $(\mu g/1)$ (µg/1)  $(\mu g/1)$ PHENOLICS Phenol . 25 20 2 4-Nitrophenol 40 N.D. 5 2-Nitrophenol 1,780... N.D. 566 2,4,6-Trichlorophenol 44 N.D. N.D. Pentachlorophenol 53 115 41.2

### PHTHALLATE ESTERS

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\*Note: Total Cyanides expressed in mg/1.

N.D. - Not Detectable, or less than detectable limits \*\*Key:

N.A. - Not Applicable

- N.S. - No Standard Available

N.P. - No Procedure Available

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TABLE	4.	(Continued	)
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Priority Pollutant	Sparged Air, (ug)	XAD-2	Sparged Air, Tenax (µg)		
POLYNUCLEAR AROMATICS					
Naphthalene	5,910			31	
2-Chloronaphthane	4,410				
Acenaphthalene	4,890				
Acenaphthene	7,590				
Fluorene	<7				
Phenanthrene/Anthracene	<6			•	
Fluoranthene	. 10				
Pyrene	57	•			
1,2-Benzanthracene	<3				
Chrysene	<3		•		
3,4-Benzopyrene	<18		•		
1,2:5,6-Dibenzanthracene	<38				
PHENOLICS					
2-Chlorophenol	1,060			•	
2-Nitrophenol	360				
Pheno1	2,400				
2,4-Dimethylphenol	304		•		
2,4-Dimethylphenol 2,4-Dichlorophenol	13,300				
2,4,6-Trichlorophenol	530				
4-Chloro-m-cresol	1,880				
2,4~Dinitrophenol	366				
4,6-Dinitro-o-cresol	507	_		•	
Pentachlorophenol	225				
4-Nitrophenol					
PURGEABLES			T <sub>1</sub>	т2	
Methylene chloride			.7	.7	
1,1-Dichloroethane			<.01	14	
<pre>1,2-Trans-dichloroethylene Chloroform</pre>			8	<.25	
1,2-Dichloroethane			<.1	<.1	
1,1,1-Trichloroethane	_		.7	<.1	
Carbon tetrachloride			<.05 380	48	
Dichlorobromomethane				<.1	
1,2-Dichloropropane			.6 .4	<.05 <.08	
Benzene	•		<.025	<.025	
Trichloroethylene					
Chlorodibromomethane			<.01	<.01	
1,1,2-Trichloroethane			<.01	<.01	
Methyl bromide		•	<.05	<.05 <.25	
Bromoform	;		<.25 17	2	
1,1,2,2-Tetrachloroethane				<del>-</del>	
Tetrachloroethylene	•		1.1	<.025	
Toluene	•		.06	.4 <.01	
Chlorobenzene			.08 <.01	.07	
Ethylbenzene			<.01	.03	

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# Plant 2

Wastewater Treatment System --

A flow diagram of Plant 2's biological wastewater treatment system is shown in Figure 5. The biological system has been in operation since August, 1971, and has incorporated powdered activated carbon (PAC) in the mixed liquor since May, 1977. Stormwater meeting permit limits is combined with final effluent, both of which ultimately enter an estuary.

Specifically, Plant 2 process wastewater receives primary clarification in an API separator, is neutralized, and is then fed to an equalization basin having a volume of 2.1 million gallons (mg) and a retention time of approximately 2 days. Wastewater in the equalization basin is lifted to two identical activated sludge bays employing surface-type extended aeration and incorporating PAC added on a batch basis to maintain an estimated 1,800 pounds of PAC in both basins. The operational result is 3,500 milligrams per liter (mg/l) of mixed-liquor volatile suspended solids (MLVSS), of which approximately 5-6 percent is PAC.

The total volume of the aeration basins is 1.1 million gallons, and the retention time is 31 hours. Mixed liquor flows by gravity to one final clarifier having a diameter of 50 feet and a retention time of 3.4 hours.

Sample Collection --

Colon Statement

The study period at Plant 2 was from July 31-August 3, 1978. Based on screening conducted under the direction of EPA's Effluent Guidelines Division, a list of priority pollutants and two common wastewater parameters (total cyanide and total phenols) was compiled for investigation in this study (Table 7). Three-liter samples were composited in 1-gallon glass bottles for analyses of specific organic compounds (no preservative added). Samples of approximately 1-liter volume were composited for the total metals (acid preservation), total cyanide (alkaline preservation), and total phenols (acid preservation).

Referring to Figure 5, the bioinfluent (point 1) was collected from a tap on the discharge side of the lift pump transferring equalization pond wastewater to the aeration basins. Return sludge (point 2) was sampled from a tap on the discharge side of the return sludge pump. Final effluent (point 3) was sampled at the weir used in monitoring for permit parameters (prior to introduction of stormwater). An air-stripper sampler equipped with both XAD and Tenax traps was floated in a corner of the south aeration bay opposite to the point where bioinfluent and return sludge are introduced. Volatile organics were sampled in the bioinfluent and final effluent using the standard 28-ml VOA (volatile organics analysis) septum vials.

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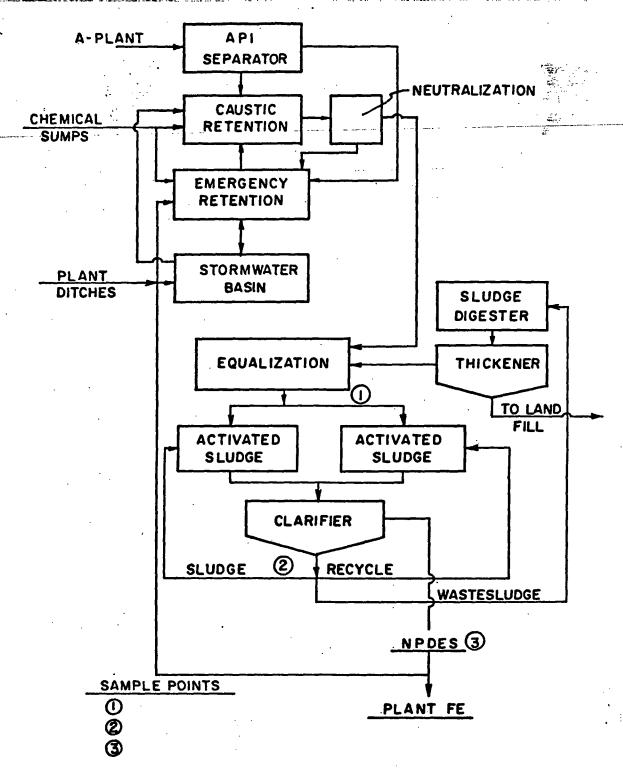


Figure 5 - WASTEWATER TREATMENT SYSTEM - PLANT 2

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Samples of bioinfluent, return sludge, and final effluent were composited every 3 hours, beginning at 6 pm on July 31, 1978, and ending at 3 pm on August 3, 1978. Preservatives were added at the initiation of sampling. Preservatives used were concentrated phosphoric acid to achieve a final pH \leq 4 (total phenols), 50-percent sodium hydroxide solution to pH \geq 12 (total cyanide), and glass-redistilled nitric acid to pH \leq 2 (total metals). The compositing procedure for each of these parameters was 40 ml of grab sample every 3 hours to obtain 1 liter of composite. For organics, no preservative was used, and final sample of 3 liters was obtained by compositing 125 ml every 3 hours. All sample containers were iced throughout the period of compositing. A detailed sampling schedule is presented in Table 5.

TABLE 5. SAMPLING SCHEDULE (PLANT 2)

Date	Tir		Sample take		Remarks	
7/31/78		30 pm			Air stripper	on
_		pm mq	Composite			
;	9 <u>1</u>	pm	**		•	
	12 i	Midnight	- "	•	·	•
8/1	.3 €	am	11			
		am	11			
		am	, 11		Forward flow	stopped
		Noon	11	•	Forward flow	resimed
	٠ _	pm	Tenax		101 101 1101	1 CD GMCG
· · · · · · · · · · · · · · · · · · ·	_	pm Pm	Composite			
		om .	11			
		o <b>m</b>	11			•
•		ym Midnight	. 11			
		. :				
8/2	3 . 8	am	11	•		
	6 a	am	11		Noted aniline influent	present in bio
•	9 a	am	11		THI TUCHL .	
· , .		Noon	11	:		
	_		Tenax			
, ,		om om	Composite			•
		•	Composite			
•	_ •	om	11			
	- I	om Kadada da ba	. 11		•	
•	12 1	Midnight	•			
8/3	2 ε	am	**		Air stripper	off
• -	3 8		Composite			
•	_	am	* 11	;	•	
:	_	am	11			
		Noon	Composite;	VOA'	S	
•	_	om	Composite	, 0.1	_	

The air-stripper sampler was placed in operation at 3:30 pm, July 31, and continued uninterrupted until 2 am on August 3, for a total operating time of 58 hours. Air charged to the stripper averaged 60 cfh, and the quantity of air to the XAD-2 scrubber was 30 cfh. Duplicate Tenax traps each received a slipstream from the stripper sampler for 15 minutes on two of the three sampling days, at 2 pm on August 1 and at 2 pm on August 2. One of the Tenax traps was left with Plant 2. The VOA samples were collected at 12 noon on August 3.

Two operational anomalies were noted which could result in atypical results for Plant 2. At 9 am on August 1, the forward feed pumps to the aeration basins were shut down, resulting in noticeable reduction in final effluent flow. The pumps resumed in 2 hours, and final flow was "normal" by 12 noon. During the morning of August 2, wastewater containing a high concentration of aniline was diverted to the emergency retention basin and bled slowly to the equalization basin. The normal 24-hour composite value for aniline in the discharge from the equalization basin is 0-5 mg/l as  $C_6H_5NH_2$ . The value for the last 24-hour period of the study (August 3) was 86 mg/l, as determined by the Plant 2 laboratory.

The daily average flows for this period are found in Table 6. Sludge wastage is not normal at this facility and did not occur during the study; therefore, the forward flow to the aeration basins equals the effluent flow.

TABLE 6. DAILY FLOW DATA (PLANT 2)

Date	Return Sludge (mgd)	Effluent (mgd)	
7/31/78	0.3283	1.3450	
8/01/78	0.3283	0.9915	
8/02/78	0.3283	0.6828	-
8/03/78	0.3010	1.1199	

Analytical Results--

Priority pollutants, total cyanide, and total phenolics for which samples were collected and analyzed are presented in Table 7. All extractions and analyses of samples were conducted at RSKERL.

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Priority Pollutant	Influent (µg/1)	Return Sludge (µg/l)	Effluent (µg/1)
CLASSICAL	:	1.25 - 1.25 - 1.25 - 1.25	i.
TOTAL CYANIDES (mg/1)*	<.08	.43	.23
TOTAL PHENOL	372	1,300	263
TOTAL METALS	-	i.	:
Arsenic	60	90	60
Selenium	10	10	20
Cadmium	<1	80	<1
Beryllium	<b>&lt;</b> 5	8	<5
Copper	32	1,700	14
Antimony	<10	<10	<10
Chromium	260	31,000	61
Nickel	36	4,200	<10
Zinc	530	39,000	300
Silver	<10	<10	<10
Thallium	<10	<10	<10
Lead	27 <5	1,500 18	<10 <5
Mercury		10	
ORGANICS (GAS CHROMATOGRAPHY)	i .		•
NITROTOLUENES/NITROBENZENE			
Nitrobenzene	N.P.	N.P.	N.P.
2,4-Dinitrotoluene	n.s.	N.S.	N.S.
2,6-Dinitrotoluene	N.S.	N.S.	N.S.
POLYNUCLEAR AROMATICS	:		
Naphthalene	<10	<10	· <10
Fluorene	63	<25	<25
Phenanthrene/Anthracene	<10	320	<10
Benzo-a-pyrene	<70	<70	<70
Acenaphthene	<10	<10	<10
PURGEABLES	:	٠	
Benzene	<40	N.A.	<40
Chloroform	<10	11	<10
1,2-Dichloropropane	<10	11	<10
Ethylbenzene	<10	11	<10
Toluene	<10	. 11	<10
Chlorobenzene	<10	**	<10
1,2-Dichloroethane	<10	11	<10
1,1,2,2-Tetrachloroethane	<10	11	<10
DICHLOROBENZENES			
1,3-Dichlorobenzene	<10	<10	<10
	<10	<10 <10	<10
1,4-Dicutoropenzene		710	720
1,4-Dichlorobenzene 1,2-Dichlorobenzene	<10	<10	<10

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	TABLE 7.	(Continued)		<del></del>
Priority Pollutant		Influent (µg/1)	Return Sludge (µg/1)	Effluent (µg/1)
PHENOLICS		1		
. 2-Nitrophenol	•	5	53	62
4-Nitrophenol	to the second contract of the second contract	24	N.D.	N.D.
Phenol		20	8	2
2.4-Dinitrophenol		, 39	16	4
2.4-Dimethylphenol		35	11	22
2-Chlorophenol		<b>. 24</b>	6	24
Pentachlorophenol		14	36	27
ACRYLONITRILE	. •		-No sample-	

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\*Note: Total Cyanides expressed in mg/l.

\*\*Key: N.D. - Not Detectable, or less than detectable limits N.A. - Not Applicable

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N.S. - No Standard Available N.P. - No Procedure Available

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Tables start PACE

Priority Pollutant	Sparged (µg)	Air,	XAD-2	Sparged Air, Tenax (µg)
POLYNUCLEAR AROMATICS		<del>~~~~</del>		
Naphthalene	40			
2-Chloronaphthane	<10			• • • •
Acenaphthalene	<10			200
Acenaphthene	-··-····· <10 -			
Fluorene	<13			
Phenanthrene/Anthracene	<10			
Fluoranthene	14	-		
Pyrene	<10			
1,2-Benzanthracene	<10			
Chrysene	<10			
3,4-Benzopyrene	<35		. :	•
1,2:5,6-Dibenzanthracene	N.D.			•
·	_			
PHENOLICS	.05			
2-Chlorophenol	<25		-	
2-Nitrophenol	<25			···· ·
Pheno1	11			
2,4-Dimethylphenol	<25			
2,4-Dichlorophenol	<50			
2,4,6-Trichlorophenol	<10			•
4-Chloro-m-cresol	60			
2,4-Dinitrophenol	<100			
4,6-Dinitro-o-cresol	80		•	•
Pentachlorophenol	<25			
4-Nitrophenol	<25			
PURGEABLES				
Methylene chloride		•		Undefinable results
1,1-Dichloroethane				because of high
1,2-Trans-dichloroethylene				background inter-
Chloroform	•			ference.
1,2-Dichloroethane	:			•
1,1,1-Trichloroethane				
Carbon tetrachloride				
Dichlorobromomethane				
1,2-Dichloropropane				
Benzene	,		•	
Trichloroethylene			•	
Chlorodibromomethane				
1,1,2-Trichloroethane				
Methyl bromide	;			
Bromoform .	i			
1,1,2,2-Tetrachloroethane				
Tetrachloroethylene	į .			•
Toluene ·	•			
Chlorobenzene	:			
Ethylbenzene				

# Plant 3

Wastewater Treatment System--

A flow diagram of Plant 3's wastewater treatment system is shown in Figure 6. Wastewaters generated by the manufacturing of organic chemicals are treated by a process consisting of (1) grit chamber, (2) primary API, (3) compositing, (4) secondary API, (5) aeration lagoon, (6) flocculation, and (7) clarification.

The typical inlet feed from the chemical sewer to the grit chamber is 1,100 gallons per minute. Wastewater flows from the grit chamber through a primary API separator into a compositing pond. The pond has a volume of 2.5 million gallons and contains two 2-horsepower mechanical agitators. Detention time is 1 day. The wastewater then flows through a secondary API separator before being pumped into a large aerated lagoon.

The aerated lagoon has a volume of  $\simeq 26$  million gallons, a depth of 22 feet, and surface area of  $\simeq 160,000$  feet. The lagoon is equipped with 32 mechanical aerators, each with a horsepower of either 50 or 75. The average detention time in the lagoon is 10 days.

Effluent from the aerated lagoon at the typical rate of 1,700 gallons per minute is pumped to a flocculator, then to a final clarifier before discharge. The final clarifier also receives wastewater from a surface sewer which has been treated by a trickling filter.

In addition to the described wastewater treatment system, this facility employs an activated sludge treatment system for treating wastewaters generated by specific process units.

Sample Collection--

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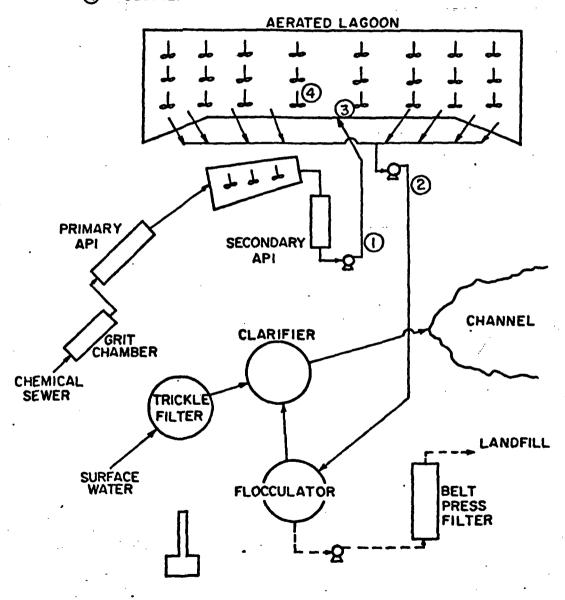
The survey of this facility was conducted during the period August 28-31, 1978. Composite samples were collected within a 72-hour period at three locations in the biological treatment system (Figure 6):

- (1) Influent to aerated lagoon (water phase)
- (2) Effluent from aerated lagoon (water phase)
- (3) Air-stripper samples (air phase)

The air-stripper sampler was located in the aerated lagoon near the influent point. The stripper sampler was placed in operation at 11 am, August 28, and operation was concluded at 7 am, August 31, for a total operating time of 68 hours. Air charged to the stripper averaged 60 cfh, and the quantity of air to the XAD-2 scrubber was 30 cfh. Two samples were collected using the stripper sampler. Stripped air was collected on the Tenax column for 10 minutes on each of the three days the sampler was operated.

# SAMPLE POINTS

- ( INFLUENT
- (2) EFFLUENT
- (3) AIR STRIPPER
- 4 SEDIMENT



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Figure 6 - WASTEWATER TREATMENT SYSTEM - PLANT 3

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Twenty-four equal aliquots were collected manually from the water-phase sampling points beginning at 12 noon, August 28. Aliquots were collected at intervals of approximately 3 hours. The final aliquot was collected at 9 am, August 31. A detailed sampling schedule is presented in Table 8.

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-- TABLE 8- - SAMPLING SCHEDULE (PLANT 3)

Date	Time	• •	Sample tal	ken		Ren	narks
8/28/79	12:00	Noon			Star Adde need	d pi	stripper at 11 am. reservative as
	3:00 6:00 9:00 12:00	pm	Composite	aliq	uot		Tenax
8/29/79	3:00 6:00 9:00 12:00 3:00 6:00 9:00 12:00	am am Noon pm pm	11 11 11 11 11 11	11 11 11 11 11 11	•	10"	Tenax
8/30/79	3:00 6:00 9:00 12:00 3:00 6:00 9:00 12:00	am am Noon pm pm	#1 #1 #1 #1 #1 #1	11 11 11 11 11 11	•	10"	Tenax
8/31/79	3:00 6:00 8:00	am ·	" " 24th"	11 11 11	•	ppei	grab, bottom sampler shut off am.

Samples were collected for those priority pollutants which were found in the previous priority pollutants screening survey or which might be expected to be found if all plant processes were in operation (Table 10). Samples were preserved by prescribed EPA methods. All samples were kept on ice throughout the sample period. Duplicate metals, cyanide, and phenolics samples were provided to plant personnel.

Park Water

At the end of the sample period, grab samples for VOA analyses were collected from the water-phase sampling points. At this time, a grab sample for the residual phase was collected from the bottom of the aerated lagoon.

Daily flow data for the sampling period are presented in Table 9.

TABLE 9. DAILY FLOW DATA\* (PLANT 3)

		T THON DITTIL (THERT S	
	Date	Influent (gpm)	Effluent (gpm)
:	8/28/79	1,100	1,700
	8/29	1,100	1,700
	8/30	1,100	1,700

\*These are typical flow rates under normal operating conditions.

Analytical Results--

THE SEASON OF THE SEASON

Priority pollutants for which samples were collected and analyzed are presented in Table 10. All extractions and analyses of samples were conducted at RSKERL.

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TABLE 10. ANALYTICAL DATA (PLANT 3)

ČENJEL. Op. 17 (1)

	!			Bottom Sediment			
din a ani	Priority P	ollutant	Influent (µg/1)	Dry Weight (µg/kg)	Effluent (µg/l)		
i	CLASSICAL	: :					
	TOTAL CYANIDES	(mg/1)*	4.76*	N.P.	6.70		
	TOTAL PHENOL		4,730	N.P.	<40		
	TOTAL METALS						
	Arsenic		17	10,000	34		
	Selenium		11	<930	20		
	Cadmium		2	670	2		
	Beryllium		<3	1,300	< 3		
	Copper		1,100	360,000	960		
	Antimony		< 10	<970	< 10		
	Chromium	,	1,400	250,000	1,400		
	Nickel		1,600	1,400,000	3,300		
	Zinc		2,000	420,000	1,900		
•	Silver		<10	<880	< 10		
	Thallium		<10	<880	< 10		
	Lead	•	380	30,000	250		
	Mercury	:	<0.1	3	<0.5		
	ORGANICS (GAS C	HROMATOGRAPHY)	•				
	ACROLEIN		<10	N.P.	<10		
	ACRYLONITRILE	<u>:</u>	<10	N.P.	<10		
	PURGEABLES	•			•		
	Benzene		<40	N.A.	<40		
	. Chloroform	ı	<10	N.A.	<10		
	1,2-Dichlo	ropropane	<10	N.A.	<10		
•	Methylene		26	N.A.	<10		
		bromomethane	N.D.	N.A.	N.D		
_		omomethane	<10	. N.A.	<10		
•	Toluene		<10	N.A.	<10		
	PHENOLICS	i					
•	Phenol	:	12	70	8		
•	2-Nitrophe	nol	11				
	4-Nitrophe		205				
70.00	2,4-Nitrop		N.D.		27		
ZGIN Astron	(Continued)				•		
LAST LEGE JE TEXT JE		; t					
	1 3 3 70"		anandan an		المراقب والمراقب والمراقب والمراقب والمراقب والمراقب والمراقب		
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TABLE 10.	(Continued)	·	
Priority Pollutant	Influent (µg/1)	Bottom Sediment Dry Weight (µg/kg)	Effluent (µg/1)
POLYNUCLEAR AROMATICS			
Benzo(a) anthracene	. <60	N.P.	<60
Benzo(a)pyrene	<30	. "	<30
3,4-Benzofluoranthene	N.D.	• •	N.D.
Benzo(k)fluoranthane	N.D.	11	N.D.
Chrysene	<20	<b>(</b> 1	<20
Acenaphthylene	<10	11	<10
Benzo(g,h,i)perylene	N.D.	11	N.D.
Fluorene	<10	**	<10
Phenanthrene/Anthracene	<10	11	<10
Dibenzo(a,h)anthracene	<30	11	<30
Indeno(1,2,3-cd)pyrene	N.D.	11	N.D.
Pyrene	126		<10
Acenaphthene	<10	11	<10
Naphthalene .	980	11 '	<10

 $\{\omega_{i,j}\}_{i=1}^{n-1}$ 

\*Note: Total Cyanides expressed in mg/l.

\*\*Key: N.D. - Not Detectable, or less than detectable limits

N.A. - Not Applicable

N.S. - No Standard Available N.P. - No Procedure Available

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TABLE 10	Sparged Air, XAD-	-2 Sparged Air, Tenax
Priority Pollutants	(μg)	-2 Sparged Air, Tenax (μg)
POLYNUCLEAR AROMATICS		
Naphthalene	· 70	
2-Chloronaphthane	16	
Acenaphthalene	<10	and the second s
Acenaphthene	<10	
Fluorene	<13	•
Phenanthrene/Anthracene	15	
Fluoranthene	50	
Pyrene	40	
1,2-Benzanthracene	<10	
Chrysene	<10	
3,4-Benzopyrene	<35	•
1,2:5,6-Dibenzanthracene	N.D.	
PHENOLICS	. 65	
2-Chlorophenol	<25	
2-Nitrophenol	<25	
Pheno1	18	•
2,4-Dimethylphenol	<25	
2,4-Dichlorophenol	<50	
2,4,6-Trichlorophenol	12	
4-Chloro-m-cresol	<25 <100	
2,4-Dinitrophenol	<25	
4,6-Dinitro-o-cresol Pentachlorophenol	<25 <25	•
4-Nitrophenol	<25 <25	
PURGEABLES		
Methylene chloride	•	Undefinable results
1,1-Dichloroethane		because of extraneou
1,2-Trans-dichloroethylene		matrix.
Chloroform		
1,2-Dichloroethane		
1,1,1-Trichloroethane		
Carbon tetrachloride		
Dichlorobromomethane		
1,2-Dichloropropane		
Benzene		
Trichloroethylene		
Chlorodibromomethane		
1,1,2-Trichloroethane	•	
Methyl bromide	•	
Bromoform	:	
1,1,2,2-Tetrachloroethane		
Tetrachloroethylene		
Toluene	1	•
Chlorobenzene	•	
Ethylbenzene		

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# Plant 4

Wastewater Treatment System --

A flow diagram of Plant 4's wastewater treatment system is shown in Figure 7. Wastewater generated by the manufacturing of pharmaceuticals is treated by a process that generally consists of (1) primary clarification, (2) equalization, (3) activated sludge, and (4) chlorine contact. Sludge produced by the wastewater treatment process is thickened, aerobically digested, and finally landfilled. In addition to the manufacturing wastewater, the treatment system also handles sanitary waste from the plant. As part of the treatment, approximately 80-100 pounds of granular activated carbon are added to the aeration basin or equalization tanks every three to four weeks.

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Treatment of the manufacturing wastewater begins when about 90,000 gallons per day of the water flows into the primary clarifier, which has a volume of 20,000 gallons (37.5 feet long by 9.0 feet wide by 7.9 feet deep). From the primary clarifier, the wastewater flows to two parallel equalization basins, each with a volume of 90,000 gallons. Then the water is combined with about 15,000 gallons per day of sanitary wastewater and flows to a 100,000-gallon aeration basin (64 feet long by 22 feet wide by 9.5 feet deep). Two-hundred twenty-five cubic feet per minute of air is supplied to the aeration basin through 18 helix air From the aeration basin, wastewater then flows to two parallel final clarifiers, each one being 26.5 feet long by 6.0 feet wide by 8.4 feet deep. The final effluent is treated in a 7.500-gallon chlorine contact tank (13.0 feet long by 13.0 feet wide by 5.9 feet deep) prior to final discharge. The total residence time of the treatment system is 2 to 2 1/2 days. sludges from the primary clarifier and waste secondary sludge are treated in an aerobic digestor and then thickened. Supernatant from the thickness is returned to the primary clarifier. sludge is then finally disposed of in a contract or municipal landfill.

Sample Collection --

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Based on the previous Effluent Guidelines Division screening survey study, the list of compounds in Table 13 was compiled. This list represents the priority pollutants that have been identified in Plant 4's influent to the treatment system. From this list, it was determined that four 1-gallon samples of composite were required for specific organic compound analysis.

The sampling point locations are shown in Figure 7. The influent sample was taken immediately preceding the primary clarifier. The effluent sample was taken just prior to chlorine contact. The return sludge sample was taken at the point where it

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## SAMPLE POINTS

- (1) INFLUENT
- 2 SLUDGE
- 3- EFFLUENT
- 4 AIR STRIPPER

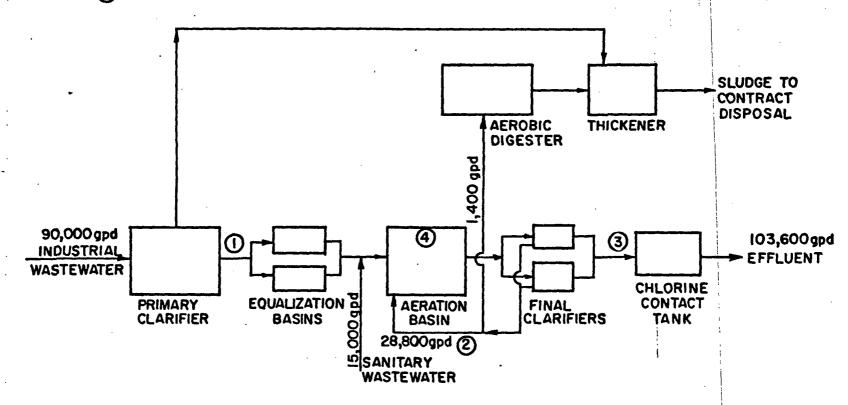


Figure 7-WASTEWATER TREATMENT SYSTEM - PLANT 4

returns to the aeration basin. The air-stripper sampler was placed in the aeration basin near the side closest to the control house.

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Samples of the influent, effluent, and return sludge were composited every 4 hours beginning at 4 pm, November 6, 1978. Four 170-ml aliquots were taken for cyanide, phenolics, and metals. Each cyanide, metals, and phenolics sample was "preserved" at about 4 pm on November 6, 1978. These samples were preserved by adding 5 ml of nitric acid to the metals samples and 2 ml of phosphoric acid to the phenolics samples. At all times each of the samples taken was kept on ice. At 8 am on November 9, 1978, VOA grab samples were taken for the influent and effluent. A detailed sampling schedule is presented in Table 11.

The air-stripper sampler was placed into operation at 10 pm on November 6, 1978, and operations were concluded at 10:30 am on November 9, 1978, for a total operating time of 58 1/2 hours. Air charged to the stripper averaged 60 cubic feet per hour, and the quantity of air to the XAD-2 scrubber was 30 cubic feet per hour. Stripped air was collected on the Tenax columns for 3 1/2 hours on November 8. Tenax sampling began at 8:40 am on November 8, 1978, and ended at 12:10 pm that day. Tenax column No. 14 was used for this sampling.

Daily flow data for the sampling period are presented in Table 12.

Analytical Results--

Priority pollutants for which samples were collected and analyzed are presented in Table 13. All extractions and analyses of samples were conducted at RSKERL.

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TABLE 11. SAMPLING SCHEDULE (PLANT 4)

Date	Time	Sample taken	Remarks
11/06/78	<u>-</u>	Composites*	Added 5 ml HNO <sub>3</sub> ,
	8:00 pm 12 Midnight	"	metals, cyanide, and phenolics composites
11/07	4:00 am 8:00 am 12 Noon 4:00 pm 8:00 pm 12 Midnight	11 11 11 11 11	phenofics composites
11/08	4:00 am 8:00 am 12 Noon 4:00 pm 8:00 pm 12 Midnight	# # # # # # # # # # # # # # # # # # #	•
11/09	4:00 am 8:00 am 12 Midnight	11 & V	OA's

\*Composites consist of four 170-ml aliquots for organics and three 50-ml aliquots for cyanides, phenolics, and metals.

TABLE 12. DAILY FLOW DATA (PLANT 4)

	TITULU TE	. 211111	I HOW DA	TU (TTUTIT A)		
	Date			Influent* (gpd)	Effluent (gpd)	
· :	11/6/78			66,910	90,640	
	11/7		:	72,450	93,850	
	11/8	•	*	71,830	92,070	
	11/9		: :	72,490	92,230	

\*Does not include sanitary wastes

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TABLE	13.	ANALYTICAL	DATA	(PLANT 4)
	~~.	MINDLITCH	NUTU	(ILUMII 7)

	Priority Pollutant	Influent (µg/1)	Return Sludge (µg/1)	Effluent (µg/1)
] ; , , · 	CLASSICAL			
	TOTAL CYANIDES (mg/1)*	<.05 *	··· <.05*	<.05
	TOTAL PHENOL	350	74	18
	TOTAL METALS			
	Arsenic	<10	<10	<10
	Selenium	<10	33	<10
	Cadmium	2	. 59	<1
	Beryllium	3	<3	<3
	Copper	120	8,300	29
	Antimony	<10	<10	<10
	Chromium	12	740	11
	Nickel .	<10	110	12
	Zinc	620	7,700	260
	Silver	<10	160	<10
	Thallium	<10	<10	<10
	Lead	12	810	<10
	Mercury	<0.8	62	<0.5
	ORGANICS (GAS CHROMATOGRAPHY)			
	PURGEABLES			
	1,2-Dichloroethane		N.A.	
	Toluene	<10	N.A.	<10
	Chloroform	<10	N.A.	<10
	Methylene chloride	<10	N.A.	48
	Benzene	<40	N.A.	<40
	Ethylbenzene	<10	N.A.	<10
	Tetrachloroethylene	10	N.A.	<10
	Trichloroethylene	<10	N.A.	<10
	PHENOLICS			
	Phenol	17	9	19
	Pentachlorophenol	18	38	26
	PHALLATES			
	Bis(2-ethylhexyl) phthallate Di-n-butyl phthallate			

\*Note: Total Cyanides expressed in mg/1.

N.D. - Not Detectable, or less than detectable limits N.A. - Not Applicable \*\*Key: TENT ENTE

N.S. - No Standard Available

N.P. - No Procedure

<u> Magn</u>

Priority Pollutant	Sparged Air, XA	D-2 Sparged Air, Tenax <sub>3</sub> 55 ft µgrams per ft
POLYNUCLEAR AROMATICS	h Prams her Til	ere herama her II
Naphthalene	90	
2-Chloronaphthane	<10	
Acenaphthalene	<10	
Acenaphthene	<10	
Fluorene	<13	
Phenanthrene/Anthracene	<10	
Fluoranthene	<10	
Pyrene	<10	
1,2-Benzanthracene	<10	
Chrysene	<10	
3,4-Benzopyrene	<35	
1,2:5,6-Dibenzanthracene	N.D.	
PHENOLICS		
2-Chlorophenol	<25	
2-Nitrophenol	<25	
Phenol	<10	
2,4-Dimethylphenol	<25	•
2,4-Dichlorophenol	80	
2,4,6-Trichlorophenol	<10	
4-Chloro-m-cresol	110	
2,4-Dinitrophenol	100	
4,6-Dinitro-o-cresol	<25	
Pentachlorophenol	<25	
4-Nitrophenol	<25	
PURGEABLES		
Methylene chloride	•	No results - sample
1,1-Dichloroethane		lost. Analytical
1,2-Trans-dichloroethylene		equipment malfunction
Chloroform	• .	
1,2-Dichloroethane		•
1,1,1-Trichloroethane		
Carbon tetrachloride		
Dichlorobromomethane		
1,2-Dichloropropane		
Benzene		
Trichloroethylene		•
Chlorodibromomethane		
1,1,2-Trichloroethane		
Methyl bromide		
Bromoform	•	
1,1,2,2-Tetrachloroethane		
Tetrachloroethylene Toluene	•	
Chlorobenzene		
Ethylbenzene Ethylbenzene		
Linyidencene		

## Plant No. 5

Wastewater Treatment System--

A flow diagram of Plant 5's wastewater treatment system is shown in Figure 8. Sanitary wastes from the plant flow into a grinder, then into a clarifier and digestor. After treatment in a chlorination basin, these wastes are added to the process stream just after the neutralization basin.

Process wastes from the manufacturing first flow into an equalization basin, from which they enter a neutralization basin of about 19,400 gallons. Acid and alkali are added as needed to adjust pH prior to the flow entering the sedimentation basin. Wastes from the sanitary system enter between the neutralization basin and the sedimentation basin.

The sedimentation basin effluent is fed to a roughing trickling filter, then into an activated sludge system which consists of two tanks in series which have a combined volume of 240,000 gallons. Feed for the activated sludge unit comes from skimmings from the dissolved air flotation unit (DAF), which is in line just after the activated sludge tank. At present, none of the skimmings are wasted. From the DAF unit, the waste is pumped by a lift station into two parallel final trickling filters of 73,500 gallons each. After treatment by the trickling filters, the waste enters a splitter and from there flows into two parallel final clarifiers. The effluent from these clarifiers is combined, then mixed with cooling water and surface runoff before discharge to the receiving stream. The sludge from the final clarifiers is dewatered by centrifugation, and the solids are incinerated.

The flow through the treatment system averages about 1.2 to 1.4 million gallons a day. The treatment effluent is mixed in a ratio of approximately 1:9 with cooling water before discharge.

Sample Collection --

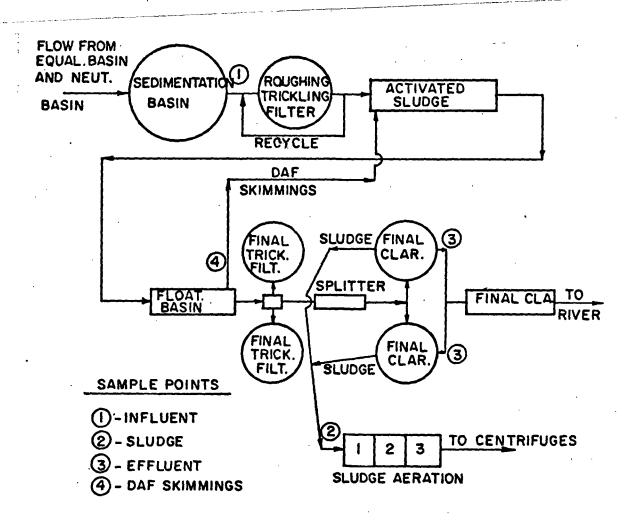
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The survey of Plant 5 was conducted September 11-14, 1978. Composite samples were collected within a 72-hour period at four points. At the final clarifiers, the discharge was mixed underground with no provision for sample-taking prior to the effluent's being mixed with surface and cooling water. For this reason, samples were taken at each final clarifier overflow, then mixed in equal proportions prior to compositing so as to produce one sample for the two locations. The samples taken were from:

(1) The sedimentation basin effluent.

(2) The final clarifier sludge taken from a valve on the discharge side of the pump that lifts sludge to the sludge aeration bays.

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Figure 8 - WASTEWATER TREATMENT SYSTEM - PLANT 5

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(3) The final clarifier effluents flowing into the weir trough at each final clarifier.

(4) The DAF skimmings which were taken at a valve located on the lift pump to the head end of the aeration tank.

The air stripper was placed in the sedimentation basin between the skimmer rail and the overflow weir. The stripper was started at 12 Noon, September 11, and operation was concluded at 9 am on September 14, for a total operating time of 69 hours. The system was shut down 2 3/4 hours prior to the prescribed 72-hour time period because of failure of both the vacuum pump and the water meter that measured flow into the air stripper. Stripped air was collected on the Tenax column for 15 minutes three times while the sampler was in operation.

Twenty-four equal aliquots were collected from the four sampling points, beginning at 12 Noon, September 11. Aliquots were collected at intervals of 3 hours. The final aliquot was collected at 8 am, September 14. A detailed sampling schedule is presented in Table 14.

TABLE 14. SAMPLING SCHEDULE (PLANT 5)

Date	Time	Sample tak	en	Remarks
9-11-78	12 Noon	Composite	aliquots	
	3:00 pm	-	"	
	6:00 pm	,	11	
	9:00 pm		**	
	12 Midnight	· ·	11,	
9-12-78	3:00 am	:	ff	
	6:00 am	:	tt	
	9:00 am	;	**	
•	12 Noon	•	11	
	3:00 pm		11	
	6:00 pm	:	11	
•	9:00 pm		11 ~	
	12Midnight	•	11	
9-13-78	3:00 am		11	
	6:00 am	:	11	Tenax, 15 min.
	9:00 am	Composite	& VOA gra	
	12 Noon	Composite	aliquots	<b>.</b>
	3:00 pm	•	11	
	6:00 pm	i	11	Tenax, 15 min.
	9:00 pm	•	11	
	12 Midnight		***	
9-14-78	3:00 am	•	**	
-	6:00 am	1;	•	
	8:00 am	,	**	Tenax 15 min.

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Samples were collected for the priority pollutants which were found in the priority pollutants survey or which might be expected to be found in plant process operations. The samples were preserved by prescribed EPA methods. All samples were kept on ice throughout the sampling period.

VOA samples were taken of the sedimentation basin effluent and the final clarifier effluent on September 13 at 9 am. Duplicate samples for cyanides and phenol analyses were taken from the final clarifier effluent for company personnel.

Throughout the sampling period, the underflow in the DAF unit was approximately 1.185 to 1.275 million gallons per day, while the skimmings or DAF sludge that was pumped varied between 151,910 to 179,940 gallons per day. The final clarifier sludge removed to aeration was about .078 million gallons per day, and the final clarifier effluent was 1.107 to 1.197 million gallons per day.

Daily flow data for the sampling period are presented in Table 15.

TABLE 15. DAILY FLOW DATA (PLANT 5)

		\ <u>-</u>	
DAF skimmings returned (mgd)	Influent (mgd)	Return sludge (mgd)	Effluent (mgd)
.178	1.2-1.4	.078	1.11-1.20
.152	<b>11</b>	ŧŧ	ŤĬ
.114	11	11	51
.118	11	tt .	. 11
	returned (mgd) .178 .152 .114	DAF skimmings Influent returned (mgd) (mgd)  .178 1.2-1.4  .152 "  .114 "	returned (mgd) (mgd) (mgd)  .178

Analytical Results--

A list of the priority pollutants for which samples were collected is presented in Table 16. All extractions and analyses were done at RSKERL. Results of these analyses are presented in Table 16.

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	Priority Pollutant	Influent (µg/1)	DAF Skimmings (µg/l)	Clarifier Effluent (µg/1)	Clarifi Sludge (µg/1)
N- -	CLASSICAL				
	TOTAL CYANIDES (mg/1)*	.25*	.22*	.17*	.23
	TOTAL PHENOL	945	107	39	20:
	TOTAL METALS	•			
	Arsenic	<10	16	<10	2
	Selenium	<10	< 10	<10	<1
	Cadmium	<01	11	· <1	ī
	Beryllium	3	< 3	< 3	<
	Copper	120	1,900	34	2,80
	Antimony	10	<10	<10	-,ss <1
	Chromium	•	1,500	34	2,10
	Nickel	39	42	38	3
	Zinc	41	5,600	. 9	7,80
	Silver	<10	17	<10	7
	Thallium	<10	<10	<10	5
	Lead	22	12	<10	3
	Mercury	<2.0	<10	<2.0	<10.
	ORGANICS (GAS CHROMATOGRAPHY)		•		
	PURGEABLES	!			
	Benzene	127	N.A.	<40	N.
	Chloroform	<10	11	<10	11
	Methylene chloride	47	**	<10	11
	Toluene	:<10	11	<10	•1
	Ethylbenzene	<sup>1</sup> 150	31	<10	11
	1,1,1-trichloroethane	<10	11	<10	*1
	1,2-dichloroethane	<10	11	<10	. 11
	PHENOLICS			•	
	4-Nitrophenol	N.D.	381	N.D.	N.
	2-Nitrophenol	123	387	21	N.
	PHTHALLATE ESTERS	į.			
	Bis (2-ethylhexyl) phthal	late		•	

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N.S. - No Standard Available

N.P. - No Procedure Available

TABLE	16.	(Continued)

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Priority Pollutant	Sparged (µg)	Air,	XAD-2	Sparged Air, Tenax (µg)
POLYNUCLEAR AROMATICS	<u>\r0/</u>			
Naphthalene	940			
2-Chloronaphthane	5,000			
Acenaphthalene	<60	-		
Acenaphthene	60			
Fluorene	<78			
Phenanthrene/Anthracene	150			
Fluoranthene	<60			
Pyrene	. <60			
1,2-Benzanthracene	<60		•	
Chrysene	<60			
3,4-Benzopyrene	<210	•		
1,2:5,6-Dibenzanthracene	N.D.	_		
	200	•		
PHENOLICS				
2-Chlorophenol	<150			
2-Nitrophenol	<150			
Phenol	<60			
2,4-Dimethylphenol	<150			•
2,4-Dichlorophenol	1,500			
2,4,6-Trichlorophenol	<10			
4-Chloro-m-cresol	1,800			
2,4-Dinitrophenol	<600		·	
4,6-Dinitro-o-cresol	600			
Pentachlorophenol	<150			
4-Nitrophenol	<150			
PURGEABLES				
Methylene chloride				Sample lost -
1,1-Dichloroethane				Analytical apparatus
1,2-Trans-dichloroethylene				malfunction.
Chloroform				
1,2-Dichloroethane				
1,1,1-Trichloroethane				·
Carbon tetrachloride				
Dichlorobromomethane				
1,2-Dichloropropane	:			
Benzene				
Trichloroethylene				
Chlorodibromomethane	٠		-	
1,1,2-Trichloroethane				
Methyl bromide				
Bromoform				
1,1,2,2-Tetrachloroethane				
Tetrachloroethylene				
Toluene	: •			•
Chlorobenzene				
Ethylbenzene				

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## Plant 6

Wastewater Treatment System--

A flow diagram of Plant 6's wastewater treatment system is shown in Figure 9. Wastewater generated by the manufacturing of pesticides and organic chemicals is collected centrally after several streams have been pretreated. The central system consists of (1) neutralization, (2) equalization, (3) activated sludge, (4) followed by a polishing lagoon. Excess sludge produced by the wastewater treatment process is disposed of in a sludge pit with a metal process waste solid. The supernatant from the sludge pit is chemically oxidized and fed to the feed of the central organic wastewater system at the pump station.

Approximately 1.7 mgd of process water from the manufacturing process is pumped to a 2.1-million-gallon equalization basin. The wastewater is mixed with return sludge from the activated sludge clarifier. This mixture is split and flows to two parallel 2.4-million-gallon aeration basins. Each basin has five 50-horsepower mechanical aerators. The mixed liquor flow from the two aeration basins is combined and flows to a single 250,000-gallon clarifier. This system operates with a return sludge flow rate of about 750 gallons per minute (gpm) and a sludge wasting rate of about 3 gpm. Figure indicates the average flow through the activated sludge unit for a 6-month period in 1978. Overflow of treated water from the clarifier flows through the polishing lagoon for final solids removal and then is mixed with combine non-contact and inorganic process water stream before discharge into San Jacinto Bay.

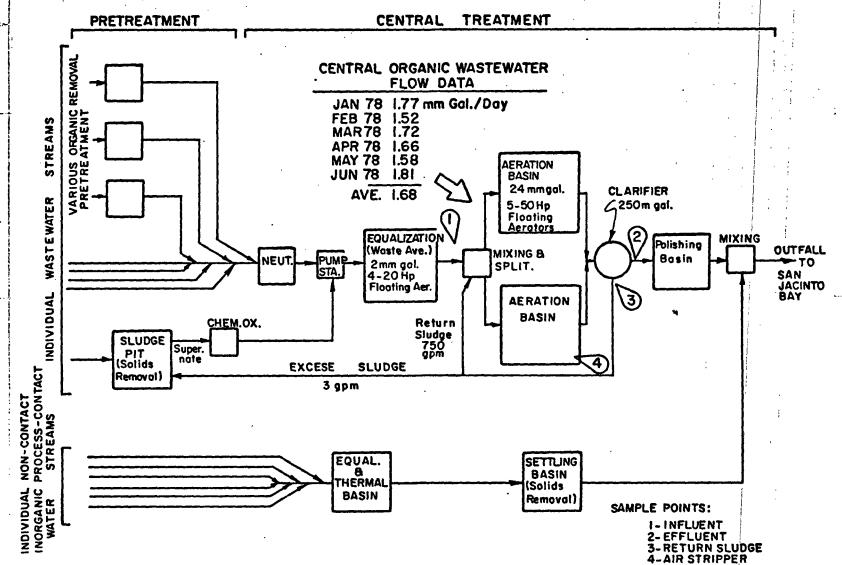
Sample Collection --

Based on the previous screening survey studies of the Effluent Guidelines Division, the list of 24 compounds in Table was compiled. This list represents the priority pollutants that were anticipated to be in Plant 6's influent to the treatment system. From this list of 24 compounds this sampling survey made quantitative determinations of 10 compounds and had trace or qualitative indication of 8 others in the influent stream. Of the 10 compounds quantitatively determined in the influent, only 5 remained with quantitative values in the effluent stream. It was determined that four 1-gallon samples of composites were required for specific organic analysis.

The locations of the sampling points are shown in Figure The influent sample (No. 1) was taken from the channel leading from the equalization basin just before the return sludge is mixed with the influent. The effluent sample (No. 2) was taken from the overflow channel of the clarifier. The return sludge

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Figure 9. WASTEWATER TREATMENT SYSTEM - PLANT 6

sample (No. 3) was taken from the sump for the recycle pump. The air-stripper sampler (No. 4) was placed in the southeast corner of the south aeration basin.

Samples of the influent, effluent, and return sludge were composited over a period of about 3 days. Every 3 hours beginning at 9 am on the first day, July 25, 1978, four 130-ml aliquots were taken at each sample point for specific organics analysis. Then, beginning at 8 am on the second day, July 26, composites were taken again every 3 hours. Finally, beginning at 7 am on the third day, July 27, composites were taken about every 3 hours. In addition, three 40-ml aliquots were taken for metals, cyanides, and phenolics at the same time as the organic samples were taken. These samples were preserved by adding 5 ml of nitric acid to the metals samples, 2 ml of sodium hydroxide to the cyanides samples, and 2 ml of phosphoric acid to the phenolics samples. At all times, each of the samples was kept on ice. At 2 pm, July 27, VOA samples were taken for the influent and effluent. A detailed sampling schedule is presented in Table 18.

The air-stripper sampler was placed into operation at 10 am on July 25, 1978, and operations were concluded at 7:30 pm, July 27, for a total operating time of 57 hours and 20 minutes. Air charged to the stripper averaged 60 cfh, and the quantity of air to the XAD-2 scrubber was 30 cfh. Stripped air was collected on the Tenax columns for 10 minutes on each day of sampling. At this plant two Tenax columns were taken. One column was retained; the other column was left with the company for its own separate analysis.

Average flow data for this plant are presented in Table 17.

TABLE 17. AVERAGE FLOW DATA (PLANT 6)

	Influent	Return Sludge
Date	(mgd)	(mgd)
1/78	1.77	_
2/78	1.52	-
3/78	1.72	-
4/78	1.66	-
5/78	1.58	-
6/78	1.81	-
Average	1.68	750

Analytical Results--

Priority pollutants for which samples were collected and analyzed are presented in Table 19. All extractions and analyses were conducted at RSKERL.

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	TABLE 18.	SAMPLING SCHEDUI	LE (PLANT 6)
Date	Time	Sample taken	Remarks
7/25/78	9:00 am	Composites	
	12 Noon	11 11	
	1:00 pm	•••	Added 5 ml HNO <sub>3</sub> , 2 ml NaOH,
	,	•	2 ml H <sub>3</sub> PO <sub>4</sub> to metals,
			<b>.</b>
a sala salah san san	3:00 pm	Composites	cyanides, and phenolics.
	6:00 pm	11	
	9:00 pm	• ·	•
7/26/78	12 Midnight	11	
, - , , , ,	3:00 am		
	6:00 am		
	8:00 am	11	
	11:00 am	11	
	2:00 pm	11 11	•
	5:00 pm	11	
•	8:00 pm		
7/27/78	11:00 pm 2:00 am	11	
1/2///0	5:00 am	11	
	7:00 am	11	
	10:00 am	11	
	1:00 pm	11	
	2:00 pm	VOA's	•
	4:00 pm	Composites	
	5:30 pm	· Effluent grabs	•
	7:00 pm	Composites	
	10:00 pm	11 11	
	11:00 pm 12 Midnight	11	

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	TYPING GUIDE SKEET	CELUSA - CELUSA		
	TABLE 19.	ANALYTICAL DATA	(PLANT 6)	
•	Priority Pollutant	Influent (µg/1)	Return Sludge (µg/1)	Effluent (µg/1)
	CLASSICAL	<u>V-61-27</u>	<u> </u>	<u> </u>
		26+	204	
	TOTAL CYANIDES (mg/1)*	.26*	.28*	.13*
	TOTAL PHENOL	190	198	273
	TOTAL METALS			
	Arsenic	20	50	20
. ,	Selenium	<10	<10	<10
	Cadmium	<1	5	<1
•	Beryllium	<5 120	30	<5
	Copper	130	36,000	120
	Antimony	<10	<10	<10
	Chromium Nickel	88 23	12,000	48
	Zinc	130	1,400	13
•	Silver	<10	5,800 <10	110
	Thallium	<10	<10	<10
-	Lead	<10	590	<10 <10
•	Mercury	<2.0	8.0	<2.0
-	ORGANICS (GAS CHROMATOGRAPHY)			
S. F				•
E	PURGEABLES	4/0	37 A	
	Benzene Chloroform	<40	N.A.	<40
ir.	•	2,240		<10
	Methylene Chloride Bromoform	10,400	11	<10
ì	Toluene	56 27		<10
C . Acti	Trichloroethylene	37	,, H	<10
	Chlorobenzene	1,620 N.D.	#1	<10
ì.	1,1,1-trichloroethane	<10	11	N.D. <10
<u>,</u> [	1,2-dichloroethane	<10	. 11	<10
	1,1,2,2-tetrachloroethane	<10	11	<10
G.	POLYNUCLEAR AROMATICS	1		110
<b>6.</b> 7	Naphthalene	110	.10	.10
e 6 l <u>é</u>	2-Chloronaphthalene	<10	<10	<10
15 1-	Benzo(a) anthracene	<10	<10	<10
	Benzo(a) pyrene	<10	<10	<10
k.	3,4-Benzofluoranthene	<43	<44	<22
	Benzo(k) fluoranthene	N.D.	N.D.	N.D.
; F	Chrysene	N.D. <10	N.D. <10	×.D.
iles:	Acenaphthylene	<10	<10	<10
į	Anthracene/Phenanthrene	<10/<10	<10/<10	<10/<10
<u>.</u>	Benzo(g,h,i)perylene	N.D.	N.D.	N.D.
	Fluorene	<16	N.D. <16	<10
<b>,</b>	Dibenzo(a,h)anthracene	<92	<94	<48
	Indeno(1,2,3-cd)pyrene	N.D.	N.D.	N.D.
	Pyrene	<10	₹.D. <10	<10
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TABLE 19. (Continued)

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Priority Pollut	ant		Influent (µg/1)	Return Sludge (µg/l)	Effluent (µg/l)
PHENOLICS	and the second of the second o	·			
Phenolics Phenol			23	12	1.3
; 					
PESTICIDES Lindane			<10	105	<10

CHART

\*Note: Total Cyanides expressed in mg/1.

\*\*Key: N.D. - Not Detectable, or less than detectable limits

N.A. - Not Applicable

N.S. - No Standard Available

N.P. - No Procedure Available

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TABLE 19.	(Continued)
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Priority Pollutant	Sparged Air, (µg)	XAD-2	Sparged Air, Tenax (µg)
POLYNUCLEAR AROMATICS			
Naphthalene	.12		
2-Chloronaphthane	<5		•
Acenaphthalene	7		
Acenaphthene	<3		
Fluorene	<13		
Phenanthrene/Anthracene	10		
Fluoranthene	. 5		
Pyrene *	<4		
1,2-Benzanthracene	<6		
Chrysene	<4		
3,4-Benzopyrene	<35		•
1,2:5,6-Dibenzanthracene	N.D.		
•	N.D.		
PHENOLICS			
2-Chlorophenol	1,380		
2-Nitrophenol	250		
Pheno1	520		•
2,4-Dimethylphenol	500		
2,4-Dichlorophenol	340		
2,4,6-Trichlorophenol	3,200		
4-Chloro-m-cresol	240		
2,4-Dinitrophenol.	<.1	•	
4,6-Dinitro-o-cresol	100		•
Pentachlorophenol	290		
4-Nitrophenol	<.02		
PURGEABLES			
Methylene chloride			
1,1-Dichloroethane			No results available
1,2-Trans-dichloroethylene			because of instru-
Chloroform			ment malfunction.
1,2-Dichloroethane			
1,1,1-Trichloroethane	•	<i>:</i>	•
Carbon tetrachloride			
Dichlorobromomethane	:		
1,2-Dichloropropane			
Benzene	•		
Trichloroethylene			
Chlorodibromomethane			
1,1,2-Trichloroethane	4		
Methyl bromide		,	
Bromoform	;		
1,1,2,2-Tetrachloroethane			
Tetrachloroethylene	•		
Toluene	<u>:</u>		
Chlorobenzene	1		•
Ethylbenzene			

## Plant 7

Wastewater Treatment System--

A flow diagram of Plant 7's wastewater treatment system is shown in Figure 10. Wastewaters generated by the manufacturing of agriculturing chemicals, including pesticides, are treated by a process consisting of (1) lime precipitation, (2) equalization, (3) acid/base neutralization, (4) activated sludge treatment, and (5) polishing in an aerated lagoon. Normally the activated sludge unit is operated without wasting sludge.

1 7 20 2

The average wastewater flow entering the treatment system is 200 gpm. The flow may be split to enter two parallel equalization basins. Normally equalization basin No. 1 receives 134 gpm, and equalization basin No. 2 receives 66 gpm. The average detention time for each basin is 3 days. Flows from the two basins are recombined for acid/base neutralization prior to being fed into the aeration basin of the activated sludge unit.

The aeration basin has a volume of 956 gallons, a depth of 15 feet, and a surface area of 16,800 feet. The basin is equipped with five 75-hp mechanical aerators. Each aerator is rated for an oxygen transfer rate of 240 pounds per hour. The average detention time in the basin is 3 days.

The final clarifier is 30 feet in diameter and 12 feet deep, with an average detention time of 5 hours. The treated water flows from the clarifier to an aerated polishing lagoon containing five 30-hp mechanical aerators prior to final discharge. Little additional biological removal as measured by TOC occurs in this lagoon. The average detention time in the lagoon is 13 days.

In addition to the described wastewater system, this facility employs two deep-well systems for disposal of wastewaters from specific process units. The plant currently produces only two pesticide products; a third pesticide producing unit has been idle since October, 1977. Wastewater from only one of the operating pesticide units goes to the biological treatment system; wastewater from the other unit goes into one of the deep-well systems.

Sample Collection--

The survey of this facility was conducted during the period June 27-30, 1978. Composite samples were collected within a 72-hour period at four locations in the biological treatment system (Figure 9):

- (1) Influent to aeration basin (water phase)
- (2) Effluent from final clarifier (water phase)(3) Return sludge (residual phase)
- (4) Air-stripper sampler (air phase)

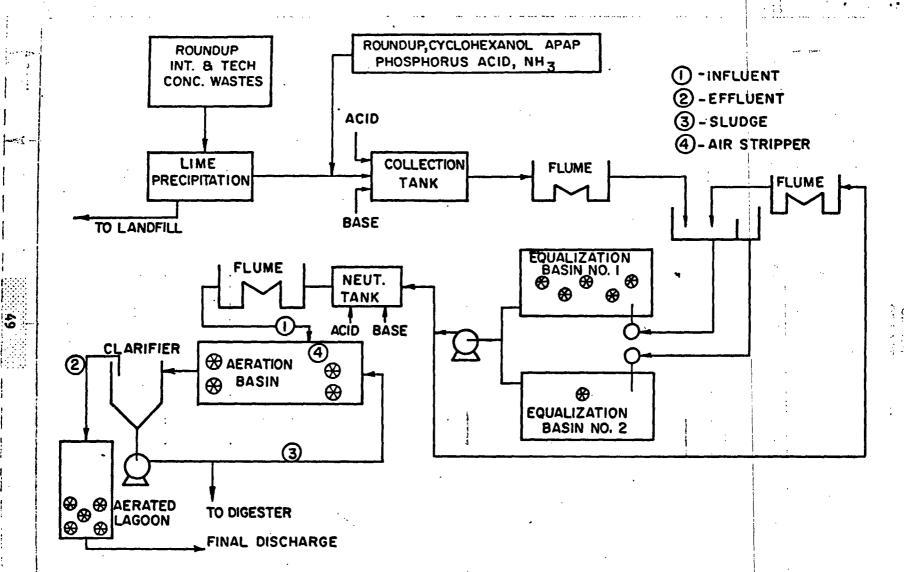


Figure 10 - WASTEWATER TREATMENT SYSTEM - PLANT 7

The air-stripper sampler was located in the aeration basin near the influent point. The air-stripper sampler was placed in operation at 8:30 am, June 27, and operations concluded at 3:30 pm, June 29, for a total operating time of 55 hours. Air charged to the stripper averaged 60 cfh, and the quantity of air to the XAD-2 scrubber was 30 cfh. Stripped air was collected on the Tenax column for 10 minutes on each of the 3 days the sampler was operating.

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TYPET TO ALBOY PROPERTY

Twenty-four equal aliquots were collected manually from the water and residual phase sampling points beginning at 8:45 am, June 27. Aliquots were collected at intervals of approximately 3 hours. The final aliquot was collected at 3 am, June 30.

Samples were collected for those priority pollutants which were found in the previous priority pollutant screening survey, or which might be expected to be found if all plant processes were in operation. Samples were preserved by prescribed EPA methods. All samples were kept on ice throughout the sampling period. At the end of the sampling period, grab samples for VOA analyses were collected from the water-phase sample points. A detailed sampling schedule is presented in Table 20.

TABLE 20. SAMPLING SCHEDULE (PLANT 7)

Date	Time	Sample taken	Remarks
6-27-78	8:45 am 12:00 Noon 3:00 pm	Composite aliquots	Tenax, 10 min.
6-28-78	6:00 pm 12:00 Midnight 3:00 am	11 11 11	each
	6:00 am 9:00 am 12:00 Noon 3:00 pm 6:00 pm	11 11 11	Tenax, 10 min.
6-29-78	9:00 pm 12:00 Midnight 3:00 am 6:00 am	11 11 11	eacii
•	9:00 am 12:00 Noon 3:00 pm 6:00 pm	VOA grab samples Composite aliquots	Tenax, 10 min. each
6-30-78	9:00 pm 12:00 Midnight 2:00 am 4:00 am	ti ti	

Duplicate sets of all samples, with the exception of the XAD slurry and Tenax columns, were collected and given to plant representatives.

Influent and effluent flow data for the sample period are presented in Table 21.

TABLE 21. DAILY FLOW DATA (PLANT 7)

	Date		Influent* (mgd)	Return Sludge (mgd)	Effluent** (mgd)
	June	27	0.25	Not Available	0.13
•	June	28	0.27	11	0.13
:	June	29	0.38	11	0.14

<sup>\*</sup>Flows entering the bio system before equalization.
\*\*Flows exiting aerated lagoon (final plant discharge).

Analytical Results--

Priority pollutants for which samples were collected and analyzed are presented in Table 22. All extractions and analyses were conducted at RSKERL.

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EPA-187 (Cm.): (4-76)

AL CYANIDES (mg/1)*  AL PHENOL  AL METALS    Arsenic    Selenium    Cadmium    Beryllium    Copper    Antimony    Chromium    Nickel    Zinc    Silver    Thallium    Lead    Mercury		.04 13,700  10 <10 2 10 40 220 51 1,600 1,500 <5 <5 5	.25 79 <10 <10 11 1 230 350 190 2 4,300 <5 <5	.04 <100 10 <10 <1 <5 <10 60 31 510 830 <5
AL PHENOL  AL METALS  Arsenic Selenium Cadmium Beryllium Copper Antimony Chromium Nickel Zinc Silver Thallium Lead		13,700  10 <10 2 10 40 220 51 1,600 1,500 <5 <5	79 <10 <10 11 1 230 350 190 2 4,300 <5	<100 10 <10 <1 <5 <10 60 31 510 830
AL METALS Arsenic Selenium Cadmium Beryllium Copper Antimony Chromium Nickel Zinc Silver Thallium Lead		10 <10 2 10 40 220 51 1,600 1,500 <5 <5	<10 <10 11 1 230 350 190 2 4,300 `<5	10 <10 <1 <5 <10 60 31 510 830
Arsenic Selenium Cadmium Beryllium Copper Antimony Chromium Nickel Zinc Silver Thallium Lead		<10     2     10     40     220     51     1,600     1,500     <5     <5	<10 11 1 230 350 190 2 4,300 <5	<10 <1 <5 <10 60 31 510 830
Selenium Cadmium Beryllium Copper Antimony Chromium Nickel Zinc Silver Thallium Lead		<10     2     10     40     220     51     1,600     1,500     <5     <5	<10 11 1 230 350 190 2 4,300 <5	<10 <1 <5 <10 60 31 510 830
Cadmium Beryllium Copper Antimony Chromium Nickel Zinc Silver Thallium Lead		2 10 40 220 51 1,600 1,500 <5 <5	11 230 350 190 2 4,300	<1 <5 <10 60 31 510 830
Beryllium Copper Antimony Chromium Nickel Zinc Silver Thallium Lead		10 40 220 51 1,600 1,500 <5 <5	1 230 350 190 2 4,300 `<5	<5 <10 60 31 510 830
Copper Antimony Chromium Nickel Zinc Silver Thallium Lead		40 220 51 1,600 1,500 <5 <5	230 350 190 2 4,300 `<5	<10 60 31 510 830
Copper Antimony Chromium Nickel Zinc Silver Thallium Lead		220 51 1,600 1,500 <5 <5	350 190 2 4,300 `<5	<10 60 31 510 830
Antimony Chromium Nickel Zinc Silver Thallium Lead		1,600 1,500 <5 <5	350 190 2 4,300 `<5	60 31 510 830
Chromium Nickel Zinc Silver Thallium Lead		1,600 1,500 <5 <5	190 2 4,300 `<5	31 510 830
Zinc Silver Thallium Lead		1,600 1,500 <5 <5	2 4,300 `<5	510 830
Silver Thallium Lead		1,500 <5 <5	4,300 `<5	830
Thallium Lead		<5 <5	`<5	
Lead		<5		
		E	~	<5
Mercury		<b>5</b>	110	5
:		<5.0	Sample lost	<5.0
ANICS (GAS CHROMATOGRAPHY)				
GEABLES	i	•		
Benzene	į	<40	N.A.	<40
Toluene		<10	11	<10
Ethylbenzene	1 2 1	<10 <10	11	
1,1,1-Trichloroethylene	. 1	<10 <10	11	<10 <10
•	!	120	. •	7.0
TICIDES			•	
Aldrin		<10	<10	<10
Dieldrin		<10		<10
Chlordane	}			
DDT .	1	<10	<10	<10
	<u>i</u>	<10	<10	<10
		<10	<10	<10
	• •	<10	<10	<10
	!	<10	<10	<10
		<10	<10	<10
	ļ	<10	<10	<10
	)	<10	<10	<10
<b>-</b>	;	<10	<10	<10
		<10	<10	<10
Heptachlor epoxide		<10	<10	<10
erinuad)				******
	Dieldrin Chlordane DDT 4,4'-DDT 4,4'-DDE 4,4'-DDD a-endosulfan-Alpha b-endosulfan-Beta Endosulfan sulfate Endrin Endrinaldehyde Heptachlor Heptachlor epoxide	Dieldrin Chlordane DDT 4,4'-DDT 4,4'-DDE 4,4'-DDD a-endosulfan-Alpha b-endosulfan-Beta Endosulfan sulfate Endrin Endrinaldehyde Heptachlor Heptachlor epoxide  mtinued)  52	Dieldrin       <10	Dieldrin       <10

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TABLE 2	2. (Co	ntinued)		
Priority Pollutant		Influent (µg/1)	Return Sludge (µg/1)	Effluent (µg/1)
PESTICIDES (Continued)	:			····
a-BHC-Alpha		<10	<10	<10
b-BHC-Beta		<10	<10	<10
r-BHC (lindane)-Gamma		<10	<10	<10
g-BHC-Delta		<10	<10	<10
Toxaphene				
PHENOLICS	:			
Pheno1	:	5,290	1.8	24.
4-Nitrophenol	•	103	N.D.	N.D.
2-Nitrophenol		17	2	71
2,4-Dimethylphenol	:	N.D.	N.D.	N.D.
2,4-Dichlorophenol	:	2	3	20
2,4-Dinitrophenol		36	14	96
2,4,6-Trichlorophenol		4	1	4
	:		•	
NITROSAMINES		·,		•
N-nitrosodimethylamine			so specific the	
N-nitrosodiphenylamine	i	detecto	or not available	<b>:</b>

CONTRE Carlotte 1

\*Note: Total Cyanides expressed in mg/1.

\*\*Key: N.D. - Not Detectable, or less than detectable limits N.A. - Not Applicable

N-nitrosodi-N-propylamine

N.S. - No Standard Available

N.P. - No Procedure Available

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TABLE 22. (Continued)

Mani de d L

Priority Pollutant	Sparged Air, XAD-2	Sparged Air, Tenax (µg)
POLYNUCLEAR AROMATICS		
Naphthalene	<.005	
2-Chloronaphthane	<.005	
Acenaphthalene	<.006	
Acenaphthene	.5	
Fluorene	<.013	
Phenanthrene/Anthracene	6	
Fluoranthene	2	
Pyrene	<.004	
1,2-Benzanthracene	<.006	•
Chrysene	<.005	
3,4-Benzopyrene	<.035	
1,2:5,6-Dibenzanthracene	N.D.	
	X • D •	
PHENOLICS	<u>_</u> _	
2-Chlorophenol		
2-Nitrophenol	<.02	•
Phenol	260	•
2,4-Dimethylphenol	. 8	
2,4-Dichlorophenol	, <b>39</b>	
2,4,6-Trichlorophenol	16	
4-Chloro-m-cresol	<.1	
2,4-Dinitrophenol	100	
4,6-Dinitro-o-cresol	11	
Pentachlorophenol	14	
4-Nitrophenol	<.02	
PURGEABLES		
Methylene chloride		<.02
1,1-Dichloroethane		<.01
1,2-Trans-dichloroethylene		<.25
Chloroform		1
.1,2-Dichloroethane	•	.01
1,1,1-Trichloroethane		<.05
Carbon tetrachloride	;	<.1
Dichlorobromomethane	to the first of the second	.3
1,2-Dichloropropane	;	<.01
Benzene	•	.01
Trichloroethylene		.002
Chlorodibromomethane	•	<.01
1,1,2-Trichloroethane	į	<.05
Methyl bromide		<.25
Bromoform	•	<.01
1,1,2,2-Tetrachloroethane	:	
Tetrachloroethylene	:	<.025
Toluene	i	<.05
Chlorobenzene	!	.006
	;	<.01
- Ethylbenzene		<.01

#### RUBBER INDUSTRY

#### Plant 8

Wastewater Treatment System--

A flow diagram of Plant 8's wastewater treatment system is shown in Figure 11. Wastewaters generated by the manufacturing of rubber and associated products is treated by a process consisting of (1) equalization, (2) neutralization, and (3) activated sludge treatment. Normally, the activated sludge unit is operated without wasting of sludge.

Treatment of the wastewater begins when process water from the various plant units enters the equalization basin at the design flow of 1,200 gpm. Volume of the basin is 8.0 million gallons with an average detention time of 110 hours at the design flow rate of 1,200 gpm.

From the equalization basin, wastewater is pumped to a neutralization basin for pH adjustment and addition of nutrients. An antifoaming agent is also added here if needed. Volume of the neutralization basin is 138,000 gallons with an average detention time of 2.0 hours at the design flow of 1,200 gpm.

From the neutralization basin, the wastewater is split to flow into two parallel aeration basins with a combined volume of 700,000 gallons. The design influent flow to each basin is 600 gpm. Each basin contains three 25-horsepower mechanical aerators. Average detention time in the aeration basins is 6.5 hours at the design flow rate.

Effluent from the aeration basins flows into a clarifier which has a diameter of 60 feet and a volume of 300,000 gallons. Retention time is 4 hours at the design flow of 1,200 gpm. Effluent from the clarifier merges with chlorinated effluent from the plant's sanitary waste treatment system for final discharge.

The plant has a 25-million-gallon stormwater basin to retain excess waters during heavy rainfall. Water from the basin is bled back through the treatment system.

#### Sample Collection--

 Elitabria (Cin.) ( **(** ) ( )

The survey of this facility was conducted during the period July 10-13, 1978. Composite samples were collected within a 72-hour period at four locations in the biological treatment system:

- Influent to aeration basin (water phase)
   Effluent from final clarifier (water phase)
   Return sludge (residual phase)

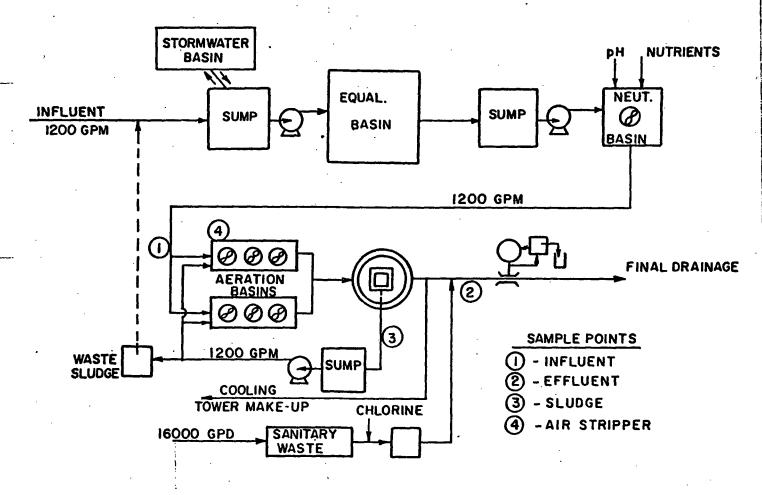


Figure 11 - WASTEWATER TREATMENT SYSTEM - PLANT 8

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The air-stripper sampler was located in one of the aeration basins as near the influent point as possible. The air-stripper sampler was placed in operation at 5:30 pm, July 10, 1978, and operation was concluded at 1:30 pm on July 13, for a total operating time of 68 hours. Air charged to the stripper averaged 60 cfh, and the quantity of air to the XAD-2 scrubber was 30 cfh. Stripped air was collected on the Tenax column for 15 minutes on each of the three days the sampler was operating.

CONTRA

Twenty-four equal aliquots were collected manually from the water- and residual-phase sampling points beginning at 5:30 pm, July 10. Aliquots were collected at intervals of approximately 3 hours. The final aliquot was collected at 2:30 pm, July 13. A detailed sampling schedule is presented in Table 23.

TABLE 23. SAMPLING SCHEDULE (PLANT 8)

Date	Time	Sample taken	Remarks
7-10-78	5:30 pm 8:30 pm 11:30 pm	Composite aliquots	
7-11-78	2:30 pm 5:30 am 8:30 am 11:30 am 2:30 pm 5:30 pm 8:30 pm 11:30 pm	11 11 11 11 11	;.
7-12-78	2:30 am 5:30 am 8:30 am 11:30 am 2:30 pm 5:30 pm 8:30 pm 11:30 pm	** ** ** ** ** ** ** ** ** ** ** ** **	
7-13-78	2:30 am 5:30 am 8:30 am 11:30 am 2:30 pm	" +	VOA grab samples

Samples were collected for those priority pollutants which were found in the previous priority pollutant screening survey or which might be expected to be found if all plant processes were in operation. All samples were preserved by prescribed EPA methods and kept on ice throughout the sampling period. At the end of the sampling period, grab samples for VOA analyses were collected from the water-phase sample points. Duplicate samples from the water- and residual-phase sample points were furnished to company personnel as requested.

Flow measurements are not taken routinely at the points where samples were collected; however, plant personnel estimated that throughput during the sampling period averaged approximately 2.4 mgd with a feed rate of 1.9 mgd and return sludge rate of 0.7 mgd.

Analytical Results--

The priority pollutants for which samples were collected and analyzed are presented in Table 24. All extractions and analyses were performed at RSKERL.

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หมลัง	TABLE 24. ANALYTICAL DATA (PLANT 8)					
DINDEPED HEND	Priority 1	Pollutant	:	Influent (µg/l)	Return Sludge (µg/l)	Effluent (µg/l)
nāgā. Patomo uz	CLASSICAL					
10.75	TOTAL CYANIDES	(mg/1)*	· · · · · · · · · · · · · · · · · · ·	<.08	<.08	<.08
	TOTAL PHENOL	:		1,190	139	24
	TOTAL METALS	•				
	Arsenic			<10	220	<10
	Selenium		į	<10	· <10	<10
	Cadmium		1	1	67	<1
	Beryllium		:	<3	16	<3
	Copper	( )	:	22	2,100	<10
	Antimony			<10	<10	<10
	Chromium		:	230	28,000	49
	- Nickel	··· 5-1-2"		89	3,500	54
	Zinc	•		1,200	150,000	180
	Silver	•		· <10	14	<10
	Thallium		:	<10	<10	<10
	Lead		į	13	2,600	<10
	Mercury			<1.5	<5.0	<5.0
	ORGANICS (GAS	CHROMATOGRAPHY)	:	ı	•	
	PURGEABLES	:	;	,		
	Benzene	• ,		<40	N.A.	<40
	Toluene	•	1	624	N.A.	<10
	Ethylbenze	ene		<10	N.A.	<10
	PHENOLICS	:			•	•
	Phenol			1	15	3
	2-Nitrophe	enol		26	52	25
	2,4-Dinita		İ	185	483	102
	4-Nitrophenol		86	485	103	
	2-Chlorophenol		N.D.	18	N.D.	
		chlorophenol		N.D.	43	N.D.
	NITROSAMINES	•				
٠	N-nitrosodiphenylamine			A nitroso specific thermal detector not available.		

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<sup>\*</sup>Note: Total Cyanides expressed in mg/l.

<sup>\*\*</sup>Key: N.D. - Not Detectable, or less than detectable limits
N.A. - Not Applicable
N.S. - No Standard Available

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OF TEXT 35-N.P. - No Procedure Available

Priority Pollutant	Sparged Air, XAD-2 (µg)	Sparged Air, Tenax (µg)
POLYNUCLEAR AROMATICS		
Naphthalene	2,560	
2-Chloronaphthane	<5	
Acenaphthalene	66	
Acenaphthene	<3	
Fluorene	<13	
Phenanthrene/Anthracene	17	·
Fluoranthene	<b>~4</b>	_
Pyrene	<4	•
1,2-Benzanthracene	<6	
Chrysene	<5	
3,4-Benzopyrene	· <35	
		·
1,2:5,6-Dibenzanthracene	N.D.	
PHENOLICS	,	
2-Chlorophenol	: <b>54</b>	-
2-Nitrophenol	1,100	
Phenol	7,800	
2,4-Dimethylphenol	2,100	• ,
2,4-Dichlorophenol	950	
2,4,6-Trichlorophenol	1,800	
4-Chloro-m-cresol	660	
2,4-Dinitrophenol	7,000	
4,6-Dinitro-o-cresol	910	
Pentachlorophenol	940	
4-Nitrophenol	1,200	
	_,	
PURGEABLES		
Methylene chloride		Lost sample -
1,1-Dichloroethane		Analytical equip-
1,2-Trans-dichloroethylene		ment malfunction
Chloroform		· matterester
1,2-Dichloroethane		
1,1,1-Trichloroethane	•	
Carbon tetrachloride	:	
Dichlorobromomethane		
1,2-Dichloropropane		
Benzene	• • • • • •	·
Trichloroethylene	:	
Chlorodibromomethane	:	
1,1,2-Trichloroethane	:	
Methyl bromide	<u>-</u>	
Bromoform	÷	
1,1,2,2-Tetrachloroethane		
Tetrachloroethylene	,	
Toluene	•	
Chlorobenzene	•	
Ethylbenzene		

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## Plant 9

Wastewater Treatment System--

A flow diagram of Plant 9's wastewater treatment system is shown in Figure 12. Wastewater generated by the manufacturing of rubber and other products is treated by a process that generally consists of (1) equalization, (2) dissolved air flotation, (3) cooling, and (4) activated sludge treatment. Normally this activated sludge unit is operated without any wasting of sludge. The plant has operated this system for a number of years and has not found a need for sludge-wasting.

CHITA

Treatment of the wastewater begins when water from the rubber plant is split into two 1,000-gpm streams and flows into two parallel equalization basins. Each basin has a 4-hour detention time at 1,250 gpm. Wastewater from the equalization basins is flash-mixed with cationic coagulant aid in a 7,500-gallon tank. The wastewater then flows to a dissolved air flotation (DAF) unit with anionic coagulant aid added as needed in line to the DAF. This DAF unit was designed by Permutit and was subsequently modified by the plant.

From the DAF unit, a portion of the wastewater (approximately 1,500 gpm) flows to a cooling tower for temperature control. After cooling, caustic, acid, and/or ammonia are added to the wastewater as required. Also, wastewaters from an oil separator and oil-skimming basin are combined with the wastewater.

Next, the wastewater flows into the aeration basin of the activated sludge unit. The aeration basin has a volume of  $1.48 \times 10^{\circ}$  gallons, a depth of 17 feet, and a surface area of  $16.870 \, \text{feet}^{2}$ . The aeration basin is equipped with four 50-horsepower mechanical aerators. These aerators are rated for an oxygen transfer rate of  $450-500 \, \text{pounds}$  per hour. The clarifier is about 85 feet in diameter and 18 feet deep. Treated water (effluent) flows from the clarifier to the bayou.

#### Sample Collection--

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Based on Effluent Guidelines Division's previous screening survey studies, the list of compounds in Table 27 was compiled. This list represents the priority pollutants that were expected to be present in Plant 9's influent to the treatment system. From this list, it was determined that four 1-gallon samples of composites were required for specific organic compound analysis.

The locations of the sampling points are shown in Figure 12. The influent sample was taken just behind the overflow weir, just before the wastewater flows into the aeration basin. The influent sample was taken off the plant's sample pump. The return sludge sample was taken off the return sludge pump. The air-stripper sampler was placed in the aerator basin just in front of the point where influent enters the basin.

80.30

37.07

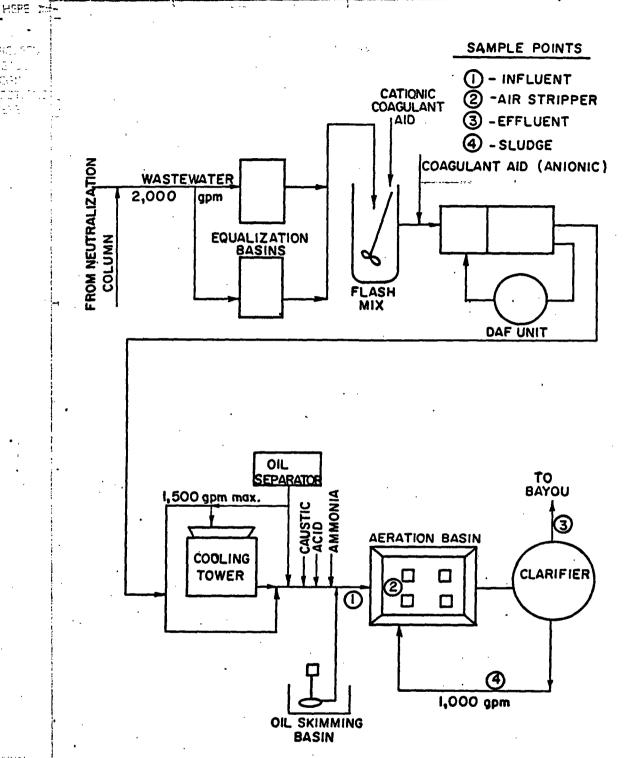


Figure 12 - WASTEWATER TREATMENT SYSTEM - PLANT 9

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LAST UND FITEKT DIS Samples of the influent, effluent, and return sludge were composited. Every 3 hours beginning at 3 pm on July 17, 1978, the first day, four 130-milliliter aliquots were taken at each sample point for specific organics analysis. Then, beginning at 2 pm on July 18, the second day, composites were taken every 3 hours. In addition, three 40-milliliter aliquots were taken for cyanide, phenolics, and metals at the same time as the organics samples were taken. Each cyanide, phenolics, and metals sample was preserved at 8 pm on July 17. These samples were preserved by adding 5 ml of nitric acid to the metals sample, 2 ml of sodium hydroxide to the cyanides sample, and 2 ml of phosphoric acid to the phenolics sample. At all times, each of the samples was kept on ice. At 7:30 pm on July 19, VOA samples were taken for the influent and effluent. I detailed sampling schedule is presented in Table 25.

TABLE 25. SAMPLING SCHEDULE (PLANT 9)

Date	Time	Sample taken	Remarks
7/17/78	3:00 pm	Composites*	
	6:00 pm	11	
	8:00 pm		Added 5 ml $HNO_3$ , 2 ml
	9:00 pm	Composites	NaOH, 2 ml $H_3PO_4$ to
	12 Midnight	•	, –
7/10/70	0.00	11	metals, cyanide, and
7/18/78	3:00 am	11	phenolics
	6:00 am	"	Pilotio 2 2 Co
	9:00 am		
	12 Noon	. 11	•
•	2:00 pm	11	
	5:00 pm	* 11	
•	8:00 pm		
	11:00 pm		·
7/19/78	2:00 am	11	
.,	5:00 am	. 11	
	8:00 am	***	
	11:00 am	11	
	1:00 pm	11	
	4:00 pm	11	
	7:00 pm	11	•
	7:30 pm	VOA's	•
•	10:00 pm	Composites	
7/20/78	1:00 am	Composites	
.,==,.•	4:00 am	11	
	-6:00 am	11 -	
	7:30 am	**	

\*Composites consist of four 130-ml aliquots for organics, three 40-ml aliquots for cyanide, phenolics and metals.

MEDIA BEAR LIMBUS

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The air-stripper sampler was placed in operation at 7 pm on July 18, 1978, and operations were concluded at 7:40 am on July 20, for a total operating time of 60 hours and 40 minutes. Air charged to the stripper averaged 60 cfh, and the quantity of air to the XAD-2 scrubber was 30 cfh. Stripped air was collected on the Tenax column for 10 minutes three times while the sampler was in operation.

Daily flow data for the sample period are presented in Table 26.

TABLE 26. DAILY FLOW DATA (PLANT 9)

			J. 1321 \	
	Date	Time	Influent (gpm)	· · · · · · · · · · · · · · · · · · ·
•	7/18/78	9:00 am	2,190	
	-7/19	9:00 am	2,220	
	7/19	6:30 pm	2,340	

Analytical Results --

Priority pollutants for which samples were collected and analyzed are presented in Table 27. All extractions and analyses of samples were conducted at RSKERL.

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Priority Pollutant	•			· · · · · · · · · · · · · · · · · · ·	T. 61	Dohum C13c-	PSE1
TOTAL CYANIDES (mg/1)*	angana Salat	Priority F	Pollutant		Influent (µg/1)		Effluen (µg/1)
TOTAL PHENOL 778 <20 <20  TOTAL METALS  Arsenic	eon. Cotiona	CLASSICAL	· :				
TOTAL METALS	91 :	TOTAL CYANIDES	(mg/1)*	• •	<.05	<.05	 <.05
Arsenic		TOTAL PHENOL	· .		778	<20	<20
Arsenic		TOTAL METALS				••	•
Selenium				:	<10	17	<10
Cadmium				•			
Beryllium				i			
Copper		•	•	;	-		
Antimony				į			
Chromium				!			
Nickel   210   36   36   36   36   36   36   36   3			•	1			
Zinc   120   250   300   Silver   10   10   10   10   10   10   10   1		- Nickel	. i 5.1/2"				
Silver			•	i			
Thallium			•	. !			
Lead   10   70   10   Mercury   <1.2   3.5   <0.5			•	!			
Mercury   <1.2   3.5   <0.5     ORGANICS (GAS CHROMATOGRAPHY)				!			
DRGANICS (GAS CHROMATOGRAPHY)   PURGEABLES   Chloroform		i					
PURGEABLES  Chloroform  Carbon tetrachloride  Methylene chloride  Toluene  Ethylbenzene  1,1,1-Trichloroethane  ACRYLONITRILE  POLYNUCLEAR AROMATICS  Phenanthrene/Anthracene  Naphthalene  Pyrene  PHENOLICS  Phenol  2,4,6-Trichlorophenol  2,4,6-Trichlorophenol  2,4,6-Trichlorophenol  2,4-Dinitrophenol  Pentachlorophenol  Pentachlorophenol  2,4-Nitrophenol  A-Nitrophenol  4-Nitrophenol  70  133  224  FIEXT P (Continued)			·HDOMATOCPADUV)				
Chloroform			MIKOHAT OGRAFITT				
Carbon tetrachloride		1			-10		
Methylene chloride		i		· i			
Toluene				İ			
Ethylbenzene			chloride				
ACRYLONITRILE   10   N.A.   10				!			
ACRYLONITRILE <10 N.A. <10  POLYNUCLEAR AROMATICS Phenanthrene/Anthracene 235 110 <10 Naphthalene 155 <10 <10 Pyrene <10 12 <10  PHENOLICS Phenol 322 1,802 7 2-Chlorophenol 7 207 207 207 207 2007 2007 2007 2007				•			
POLYNUCLEAR AROMATICS  Phenanthrene/Anthracene 235 110 <10 Naphthalene 155 <10 <10 Pyrene <10 12 <10  PHENOLICS  Phenol 322 1,802 2 2-Chlorophenol 7 207 2,4,6-Trichlorophenol 15 72 2,4-Dinitrophenol 183 N.D. 140  GIN Pentachlorophenol 252 N.D. 143  GIN Pentachlorophenol 70 133 224		1,1,1-Tric	thloroethane	į	· <10	<b>"</b>	<10
Phenanthrene/Anthracene   235   110   <10	-	ACRYLONITRILE			<10	N.A.	<10
Phenanthrene/Anthracene   235   110   <10		POLYNUCLEAR ARC	MATICS				
Naphthalene		1			235	110	<10
Pyrene   <10   12   <10				İ			
Phenol   322   1,802   2   2-Chlorophenol   7   207   2,4,6-Trichlorophenol   15   72   2,4-Dinitrophenol   183   N.D.   140	•		i				
Phenol   322   1,802   2   2-Chlorophenol   7   207   2,4,6-Trichlorophenol   15   72   2,4-Dinitrophenol   183   N.D.   140   252   N.D.   143   254   254   255   25	•	DUENOI TCC					
2-Chlorophenol   7   207   2,4,6-Trichlorophenol   15   72   2,4-Dinitrophenol   183   N.D.   140		1	i		220	1 000	
2,4,6-Trichlorophenol   15   72   1   2,4-Dinitrophenol   183   N.D.   140	-	4		•	_		
2,4-Dinitrophenol   183   N.D.   140				1	· -		
GIN Pentachlorophenol 252 N.D. 142 SITEME 4-Nitrophenol 70 133 224 TEXT IN (Continued)				1			_
TEXT IF (Continued) 70 133 224	€ INI			ĺ			
TEXT IN (Continued)				İ			
		(Continued)	HOT	1	70	133	224
	12/11/20			<del>+</del>			

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Effall OF OF CAGE (EXT HT 6 3 % TABLE 27. (Continued) Influent Return Sludge **Effluent** 0.107930  $(\mu g/1)$ Priority Pollutant  $(\mu g/1)$  $(\mu g/1)$ 125C DEC 19 DEC 1941.6 **NITROSAMINES** - 5.5 N-nitrosodiphenylamine --- A nitroso specific thermal detector not available. Total Cyanides expressed in mg/l. \*\*Key: N.D. - Not Detectable, or less than detectable limits N.A. - Not Applicable N.S. - No Standard Available N.P. - No Procedure Available 9-1/8" SEGIN LAST LINE OF TEXT IN-£ 3/3″ 66

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TABLE	27. (Continued)		<del>-</del>
Priority Pollutant	Sparged Air, XAD-2 (μg)	Sparged Air, Tenax (µg)	<del></del>
POLYNUCLEAR AROMATICS			
Naphthalene	Undefinable GC		
2-Chloronaphthane	chromatograph obtain	ned	
Acenaphthalene	because of matrix in		
Acenaphthene	ferences		
Fluorene			
Phenanthrene/Anthracene			
Fluoranthene	•		
Pyrene	•		
1,2-Benzanthracene			
Chrysene	<b>.</b>		
3,4-Benzopyrene	1		
1,2:5,6-Dibenzanthracene	•		
•			
PHENOLICS			
2-Chlorophenol	Undefinable GC		
2-Nitrophenol	chromatograph obtain	ned	
Pheno1	because of matrix i	nter- ,	
2,4-Dimethylphenol	ferences		•
2,4-Dichlorophenol			
2,4,6-Trichlorophenol			
4-Chloro-m-cresol	. !		
2,4-Dinitrophenol	· · · · · · · · · · · · · · · · · · ·		
4,6-Dinitro-o-cresol	! •		
Pentachlorophenol	<b>;</b>		
4-Nitrophenol	:		
PURGEABLES	•		
Methylene chloride	<u>:</u>	.01	•
1,1-Dichloroethane	•	.5	
1,2-Trans-dichloroethylene		.2	
Chloroform	<u> </u>	.2	
1.2-Dichloroethane	•	<.1	
1,1,1-Trichloroethane	·	<.05	
. Carbon tetrachloride		<.1	
Dichlorobromomethane		<.05	
1,2-Dichloropropane		<.01	
Benzene	· •	<.02	
Trichloroethylene		.07	
Chlorodibromomethane		<.01	
1,1,2-Trichloroethane	•	<.05	
Methyl bromide		<.25	
Bromoform -	.	<.01	
1,1,2,2-Tetrachloroethane	•	<.02	
Tetrachloroethylene		<.02 <.05	
Toluene	•	<.01	·:
Chlorobenzene	Ì	<.01	
Ethylbenzene	•	<.01	

#### WOOD-PRESERVATIVES INDUSTRY

## Plant No. 10

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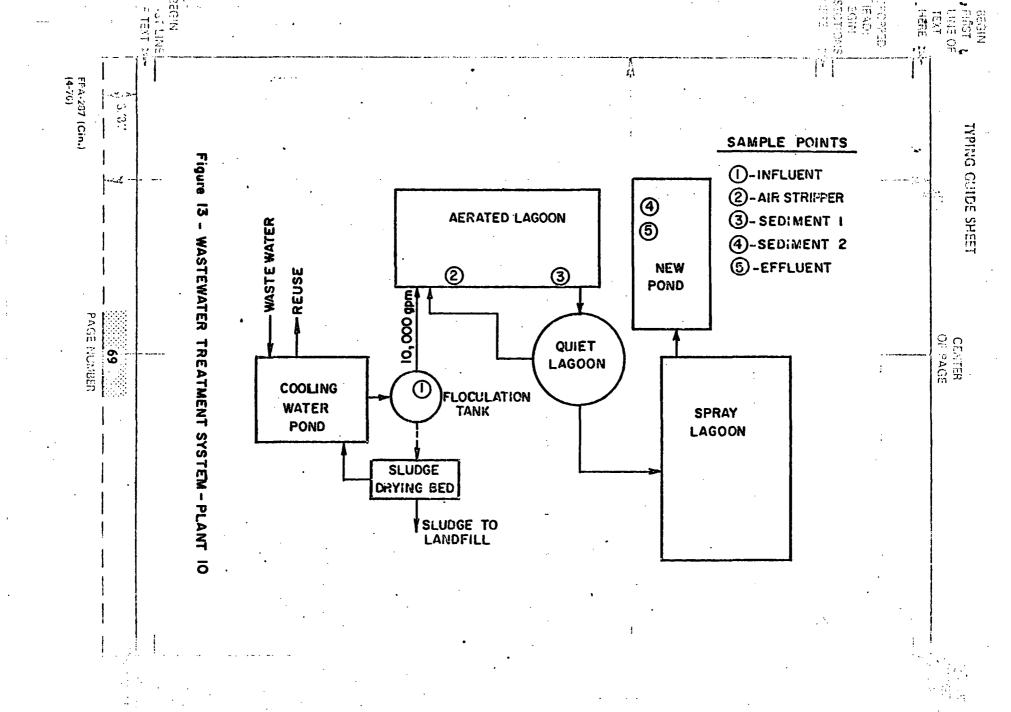
Wastewater Treatment System --

A flow diagram of Plant 10's wastewater treatment system is shown in Figure 13. Wastewater generated by the wood-treating process flows into a vacuum cooling-water pond. Cooling water for the barometric condensor is recirculated from the pond through the condensor and back to the pond. As the level of the pond rises as a result of the addition of wastewater, a portion of the water is removed and treated. The treatment system generally consists of (1) chemical flocculation, (2) nutrient addition, (3) aeration, (4) spray pond evaporation, and (5) total retention.

Treatment begins when the flocculation tank is filled with an approximately  $\bar{5}$ ,000-gatlon batch of water from the pond. A Bentonitic clay is then mixed in a drum with water and added to the tank. This is well mixed into the water, and then a cationic liquid polymer is added, which serves only to gather the floc particles together to increase their size and speed settling of the floc. The tank is allowed to settle for  $1 \frac{1}{2}$  to 3 hours, after which the clarified water is pumped from the tank to the aeration basin. The tank (7 feet high) is decanted to within only 18 inches of the bottom. The sludge from the first batch will all be below this point. The sludge is not removed at this time; instead, the tank for the next batch flocculation is refilled as needed. The sludge will react with the new floc in the same way as did the clay in the first mix; so no more clay is added, only the polymers. Again, the tank is allowed to settle. Afterwards it is decanted and remixed until the sludge is near the decant level, at which time the sludge is pumped to drying beds. The sludge has excellent drying characteristics, unlike some others, and when dry is seemingly unaffected by rainfall. When dry, the sludge from 60 to 80 thousand gallons of water will be approximately 3 cubic yards and can easily be handled for disposal at a landfill. The next step in the process is aeration and nutrient addition. Ammonium nitrate and phosphate fertilizers are added daily at the rate of approximately 20 pounds of nitrogen and 1 pound of phosphate for each 100 pounds of COD. Clarified water is transferred to a 1-milliongallon aerated lagoon.

Air is supplied by two 10-hp positive displacement blowers with a rated capacity of 150 cfm each. These blowers supply air through a pipe system at a depth of 4 feet. After the aeration basin, the effluent flows to a smaller settling pond where most of the biological sludge settles and is returned to the aeration basin. The effluent then flows to a large pond, where it is extensively sprayed into the air to evaporate it, as well as adding considerably more aeration. Following the spray lagoon, the

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wastewater finally flows to a l-million-gallon retention pond. One notable feature of this plant is the fact that it does not discharge any effluent. Normally the wastewater is disposed of entirely by evaporation.

Sample Collection--

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Fig.

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Based on the previous Effluent Guidelines Division Screening Survey study, the list of compounds in Table 30 was compiled. This list represents the priority pollutants that have been identified in Plant No. 10's influent to the aerated lagoon. From this list, it was determined that four 1-gallon samples of composite were required for specific organic compound analysis. The locations of the sampling points are shown in Figure 12. The sample of the influent to the treatment system was taken directly out of the top of the flocculation tank. The "effluent" sample was taken from the new pond (or retention pond) at a point farthest from its influent. It should be noted that this plant does not normally discharge any effluent; however, the "effluent" sample point represents the water that would be discharged if an effluent were allowed.

Two sediment samples were taken. One sample was taken from the aerated lagoon, and the other was taken from the final retention pond. The aerated lagoon sediment sample was taken near the point where the wastewater flows from this lagoon to the quiet lagoon. The final lagoon sediment sample was taken near the "effluent" sampling point. The air-stripper sampler was set up in the aerated lagoon just off the walkway located approximately 60 feet from the point where the influent enters the system.

Clarified water from the flocculation tank was added twice daily (during daylight hours) to the aeration lagoon; thus sampling times were coordinated to the operation of the flocculation tank. All liquid samples were taken during the times influent was being added to the treatment system. One-sixth of the total volume of composite required was collected for each batch of influent. Effluent samples were collected at the same time as were influent samples. A detailed sampling schedule is presented in Table 28.

The air-stripper sampler was placed in operation at 2:58 pm on June 20, 1978, and operation was concluded at 10:50 am on June 22 (for a total operating time of 12 hours). The air-stripper sampler was operated during daylight hours to correspond roughly to the liquid sampling. Air charged to the stripper averaged 60 cfh, and the quantity of air to the XAD-2 scrubber was 30 cfh. Stripped air was collected on the Tenax column for 10 minutes each day the stripper was in operation.

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FEST

ishni sebi Hawbi Daily flow data for the sample period are presented in Table 29.

TABLE 29. DAILY FLOW DATA (PLANT 10)

- Date	Influent (gal.)	Effluent (gal.)
6/20/78	10,000	0
6/21	10,000	0
6/22	10,000	0

Analytical Results--

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Priority pollutants for which samples were collected and analytical results are presented in Table 30. All extractions and analyses were conducted at RSKERL.

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TUNT		OF 1	PAGE				IMAGE ASEA
HERE 7	TABLE 30.	ANALYTIC	AL DATA	A (PLANT 1	.0)		
# · · · ·				Botto	m Sediment		•
-DROPPED HEAD		_			y Weight		
BEGIN	Budanden Ballutant		fluent		μg/kg)	Effluent	
SECTIONS	Priority Pollutant		μg/1)	Aer. Lag	. Final Pond	(µg/1)	
HEME 1	CLASSICAL		<u> </u>				
	TOTAL CYANIDE (mg/1)*		<.02*	N.P.	N.P.	<.02*	
					•	•	
•	TOTAL PHENOL	17	0,000	**		<100	
	TOTAL METALS						
	Arsenic	· (	60	9,300	7,600	20	
	Selenium		<10	<4,500	<1,600	<10	
	Cadmium		<1	2,100	<1,300	<1	
	Beryllium		<5	<1,900	3,200	<5	
	Copper		<10	40,000	4,300	<10	
	-4 -Antimony - 3-1/2" -		2	- <3 <b>,</b> 700 ····	•	<10	•
•	Chromium		16	5,600	3,100	<10	
	Nickel		30	19,000	18,000	<10	•
	Zinc		160	310,000	48,000	<10	
	Silver		<5	<2,100	<690	<5	
	Thallium Lead		8	6,300	<690	<5	
	Mercury		<5 <.5	<2,100 <136	27,000 <3.4	<5-	<b>,</b>
	Mercury	,	\.,	/120	<b>\3.4</b>	<.5	
•	ORGANICS (GAS CHROMATOGRAPHY)						
	DUD CD A DI PG			•			
	PURGEABLES Benzene		107	N D	<b>17</b> D	-10	
	Chloroform		187 <10	N.P.	N.P.	<40 -10	
	Methylenechloride	!	<10	H	11	<10 <10	
	Ethylbenzene		450	11		. <10	
	Dichlorobromomethane	. •	<10	**	**	<10	
	Toluene		300	11	n <sup>'</sup>	<10	•
				•			
	POLYNUCLEAR AROMATICS				•	;	
	Benzo(a) anthracene		<20	3,700	149	<20	
	Benzo(a) pyrene		<10	<310	<310	<10	
	3,4-benzofluoranthene		N.S.	N.S.	n.s.	N.S.	
	Chrysene		<10	4,500	2,060	<10	
	Acenaphthylene		<10	<100	<100	<10	
	Benzo(g,h,i)perylene Fluorene		N.S.	N.S.	N.S	N.S.	• •
	Phenanthrene/Anthracene	•	<10 <10	17,600	210	<10/10	
•	Dibenzo(a,h)anthracene		<10	19,500 N.D.	3,390 N.D.	<10/10 <10	
	Indeno(1,2,3-cd)pyrene	i	N.S.	N.S.	N.D. N.S.	N.S.	11 POM OF
BEGIN	Pyrene		<10	5,300	4,140	<10	WE AREA
LAST LINE	· ·		13	5,110	<250	<10	FLUFUIDE FALENSION
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	TABLE	30. (0	Continue	d)		
Pri	ority Pollutant	Sparg	ged Air, (μg)	XAD-2	Sparged Air (µg)	, Ten
POLYNUC	LEAR AROMATICS,					
Na	phthalene		286			
2-	Chloronaphthane	;	46		-	
Ac	enaphthalene		<3		and the second s	
Ac	enaphthene	•	54			
	uorene	1	28			
Ph	enanthrene/Anthracene		55			
4	uoranthene	•	<4			
Pv	rene	•	<4			
•	2-Benzanthracene	•	<28			
	rysene	:	. <9			
	4-Benzopyrene		<15			
	2:5,6-Dibenzanthracene		N.D.			
•			n.D.			
PHENOLI		•				
	Chlorophenol	390	0,000			
	Nitrophenol	63	L,000			
ī	enol	47	7,000		•	
	4-Dimethylphenol	48	3,000			
2,	4-Dichlorophenol	. 74	,000			
2,	4,6-Trichlorophenol		2,000			
4-	Chloro-m-cresol	4	,500			
2,	4-Dinitrophenol	. !	540			
4,	6-Dinitro-o-cresol	: 3	3,000			
Pe	ntachlorophenol	!	630			•
4-	Nitrophenol	· ·	5,100	•		
PURGEAB	LES					
Me	thylene chloride		•		2	
	l-Dichloroethane	1			.9	
	2-Trans-dichloroethylene				.1	
	loroform	•			.1	
	2-Dichloroethane	1			.5	
	l,l-Trichloroethane				.06	
	rbon tetrachloride	÷ :	:		3	
,	chlorobromomethane	į			2	
1	2-Dichloropropane				.2	
	nzene				3	
•	ichloroethylene	}		•	<.01	
	lorodibromomethane	i			.1	
•	l,2-Trichloroethane	1	•		<.05	
	thyl bromide	. ! .	•		<.02	
	omoform	• !				
	l,2,2-Tetrachloroethane	:			.03	
	trachloroethylene	:			<.02	
	luene	į	•		4	
1	lorobenzene	Ì			<.01	
,	hylbenzene	ì			<.01 <.01	

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# Plant 11

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Wastewater Treatment System--

A flow diagram of Plant 11's wastewater treatment system is presented in Figure 14. Wastewaters generated by a wood-preserving (creosote) plant are treated by a combined biological/land irrigation process. The process consists of (1) settling, (2) storage, (3) aerated treatment, (4) spray irrigation, and (5) runoff storage. Rainfall runoff water is collected in a storage pond, recycled through 10 spray nozzles, and then bled into the settling basin for treatment.

The average wastewater flow entering the treatment system is 50,000 gallons a day. Wastewater flows from the settling basin to a storage pond prior to entering the aerated lagoon. Detention time in the storage pond and aerated lagoon is 40 to 60 days each.

The aerated lagoon contains four 7.5-hp mechanical aerators. The wastewater is intermittently pumped from the lagoon and sprayed through 18 spray nozzles in three sections onto an eight-acre field growing fescue and sericea lespedeza grasses. Runoff from the field is collected in a runoff storage pond.

Water from the storage pond is recycled to the plant, where it is treated and used as boiler feedwater, cooling water, etc. The plant has no wastewater discharge.

Sample Collection --

The survey of this facility was conducted during the period August 7-10, 1978. Composite samples were collected at three locations in the treatment system (Figure 13):

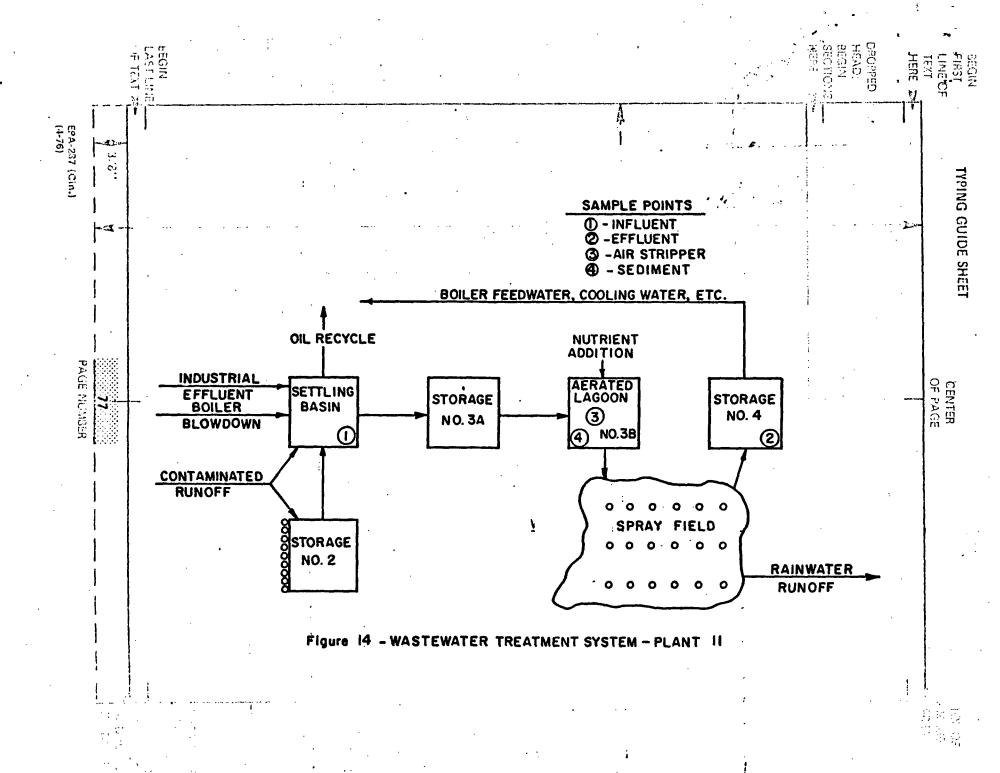
- (1) Influent to aerated treatment (water phase)
- (2) Runoff storage pond (water phase)
- (3) Air-stripper sampler (air phase)

The air-stripper sampler was located in the aerated lagoon. The air-stripper sampler was placed into operation at 2:50 pm, August 7, and operation was concluded at 6:50 am on August 10, 1978, for a total operating time of 64 hours. Air charged to the stripper averaged 60 cfh, and the quantity of air to the XAD-2 scrubber was 30 cfh. Stripped air was collected on the Tenax column for 10 minutes on each of the three days the sampler was operated.

Twenty-four equal aliquots were collected manually from the water-phase sampling points, beginning at 3 pm, August 7. Aliquots were collected at intervals of approximately 3 hours. The final aliquot was collected at 10 am, August 10. A detailed sampling schedule is presented in Table 31.

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TYPING GUIDE SHEET FIRST CENTER LINE OF OF PAGE TEXT HERE IN-DAOPPED HEAD. SAMPLING SCHEDULE (PLANT 11) TABLE 31. CEGIN SECTIONS Time Sample taken Date Remarks HE HE 8-7-78 3:00 pm Composite aliquots Rain showers 6:00 pm \*\* 9:00 pm 12 Midnight 8-8-78 3:00 am 6:00 am 9:00 am No rain 12 Noon 3:00 pm Tenax, 20 min. 6:00 pm 9:00 pm 12 Midnight 8-9-78 3:00 am 11 6:00 am 9:00 am 9-1/3:00 pm Tenax, 20 min. 6:00 pm 9:00 pm 12 Midnight 8-10-78 3:00 am Rain showers 6:00 am Tenax, 20 min. 9:00 am VOA grab samples, bottom sediment sample 110:00 am Final composite aliquot Samples were collected for those priority pollutants which were found in the previous priority pollutants screening survey. Samples were preserved by prescribed EPA methods. All samples were kept on ice throughout the sample period. At the end of the sample period, grab samples for VOA anal-

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yses were collected from the water-phase sample points. At this time, a grab sample for the residual phase was collected from the bottom of the aerated lagoon.

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Daily flow data for the sample period are presented in Table 32.

TABLE 32. DAILY FLOW DATA (PLANT 11)

Date	Influent* (mgd)	Return Sludge (mgd)	Effluent (mgd)
8-7-78	Not available for	N.A.	0.05
8-8-78	these dates	N.A.	0.05
8-9-78		N.A.	0.05

\*Average influent flow: 0.05 mgd

Analytical Results--

Priority pollutants for which samples were collected and analyzed are presented in Table 33. All extractions and analyses were conducted at RSKERL.

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.• 	TABLE 33.	ANALYTICAL	DATA

	· · · · · · · · · · · · · · · · · · ·	Bottom Sediment	•
Bud and the Dollar town	Influent	Dry Weight	Effluent
Priority Pollutant	(ug/1)	(µg/kg)	(µg/1)
CLACCICAL			!
CLASSICAL			
TOTAL CYANIDE (mg/1)*	<.08*	N.P.	<.08*
TOTAL PHENOL	79,000	N.P.	16
TOTAL METALS			
TOTAL METALS	520	51 000	∠E∩
Arsenic	530	51,000	<50
Selenium	<10	940	<10
Cadmium	< 1	200	< 1
Beryllium	< 5	2,500	< 5
Copper	44	99,000	16
Antimony	<10	<990	<10
	_ +260	56,000	<10
Nickel	22	18,000	<10 -
Zinc	. 70	280,000	100
Silver	<10	<910	<10
Thallium	<10	910	<10
Lead	<10	20,000	15
Mercury	<.5	<b>∹20</b>	<.5
			•
ORGANICS (GAS CHROMATOGRAPHY)			
PURGEABLES			
Benzene	<40	N.P.	<40
Chloroform	<10	M.P.	<10
Methylene Chloride	32	11	16
Ethylbenzene	156	11	<10
Dichlorodibromomethane	N.S.	. 19	
Toluene	31	11	N.S. <10
Totalene	31	•	<b>~1</b> 0
POLYNUCLEAR AROMATICS			i i
Benzo (a) anthracene	<20	1250	<20
Benzo (a) pyrene	<10	5980	<10
3,4-benzofluoranthene	··	N.S.	N.S.
Chrysene	67	9280	<10
Acenaphthylene	670	1400	<10
Benzo (g,h,i) perylene	N.S.	N.S.	N.S.
Fluorene	42	547	<10
Phenanthrene/Anthracene	1900	43,700	<10/10
Dibenzo (a,h)anthracene	<10	N.D.	<10
Indeno (1,2,3-cd) pyrene	N.S.	N.S.	NC
Pyrene	570	4250	10
Acenaphthene	400	1840	-10
Naphthalene	<10	<104	~10 ' '
		<b></b> ,	
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¥3.0 ¥	80		1 T

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RE	-	TABLE 33	. (Continued)			
PED D;	Priority P	ollutants	Influen (µg/l)	Bottom Sedi t Dry Weig	ht Effluent	
N OI	RGANICS (GAS C	HROMATOGRAPHY)				•
	HENOLICS	:			<del></del> . `	
	Phenol		10,900	4,500	2	
ł	2,4-dimeth 2-chloroph		N.D. 200	N.D. 300	26 N.D.	
	2,4,6-tric	hlorophenol	420	N.D.	37	
	Pentachlor	opheno1**	6,820	4,800	105	
		<u> </u>				_
	**Pentachlorop	. Cyanide expressed :	le.			
*:	**Key: N.D	Not Detectable, or	less than dete	ctable limits		
		Note Applicable No Standard Availal	ole			
		No Procedure Avail		•		· •
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HIGHE X	TAB	LE 33.	(Continue	d)	<del></del>		· · ·
RCHEO	Priority Pollutant	Spa	rged Air, (µg)	XAD-2	Sparged Air (µg)		_
e karili ili Parili ili	POLYNUCLEAR AROMATICS						-
iner Termoni	Naphthalene		4,230				
	2-Chloronaphthane	;	7			. •	
	Acenaphthalene		<3		<del></del>		
	Acenaphthene		2				
·	Fluorene	]	<5				
	Phenanthrene/Anthracene	i	7		•		•
	Fluoranthene	:	15				
	Pyrene	i	13			-	
	1,2-Benzanthracene	!	70		•		
	Chrysene	į	24				
	3,4-Benzopyrene	1	<15				
	1,2:5,6-Dibenzanthracene	i .	N.D.				
		:					
	PHENOLICS		. 551 - 1				
	2-Chlorophenol		ufficient			<del>_</del>	
	2-Nitrophenol	to	run pheno	lics			•
	Phenol	j					
	2,4-Dimethylphenol	İ					
	2,4-Dichlorophenol					•	•
	2,4,6-Trichlorophenol	!				-	
	4-Chloro-m-cresol	}			•		
,	2,4-Dinitrophenol	i		•	_	•	
!	4,6-Dinitro-o-cresol						
	Pentachlorophenol	İ					
	4-Nitrophenol		-	•			
	PURGEABLES	!					
	Methylene chloride	j			Undefinable	GC result	ts
	1,1-Dichloroethane	Ì			because of		
	1,2-Trans-dichloroethylene	:			retention t		
	Chloroform	i			ternal stan		
	1,2-Dichloroethane	{	•				
	1,1,1-Trichloroethane	į			•		
Ċ	Carbon tetrachloride						:
	Dichlorobromomethane						•
	1,2-Dichloropropane	į		•			
•	Benzene	1		1			: •
	Trichloroethylene		•	,	•		<i>†</i> :
1	Chlorodibromomethane	Ì	•				· ! !
	1,1,2-Trichloroethane						
	Methyl bromide					•	
	Bromoform	1					
	1,1,2,2-Tetrachloroethane		•				1
	Tetrachloroethylene	ì			•		I
į	Toluene	į					77 12.5
Giri	Chlorobenzene	•					
AST LILE		!					
CENT A	Eliyibenzene						(i. s)
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# Plant 12

March 1988

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Wastewater Treatment System --

A flow diagram of Plant 12's biological wastewater treatment 55.0 DOM: system including sample points is shown in Figure 15. The biological system consists of two completely separated bays receiving overflow from two dissolved air flotation units. The two bays are identical in hydraulic characteristics and operation; but bay A utilizes powdered activated carbon in the mixed liquor, while bay B does not. At the time of the study, both bays were operating at steady state. The type of biological treatment utilized in each bay is extended aeration with surface aerators. Waste sludges, including wasted return sludge, receive thickening by filtration prior to land disposal.

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Specifically, Plant 12 process wastewater (approximately 4.0 mgd) receives primary clarification with two API separators (each having a volume of 0.084 mg; dimensions, 80 feet long by 20 feet wide by 7 feet deep). The API effluents are fed to respective dissolved air flotation (DAF) units (each having a volume of 0.060 mg; dimensions, 55 feet long by 20 feet wide by 7.25 feet The DAF effluents are combined and lifted to two identical aeration bays and split equally between the bays. Each bay consists of an aeration basin and final clarifier in concentric juxtaposition, the clarifler being innermost. Each aeration basin has a volume of 2.28 mg and dimensions of 184 feet OD, 89 feet ID, and 15 feet deep. Each clarifier has a volume of 0.56 mg and dimensions of 89 feet in diameter by 12 feet deep. average daily forward flow for the study period (November 5-8, 1978) was 3.6 mgd. The return sludge pumping rate for each bay was 0.72 mgd.

Sampling Program --

Based on previous screening of the refining industry by the EPA's Effluent Guidelines Division, a list of priority pollutant compounds was compiled for investigation in this study (Table 22). In addition, two common wastewater parameters were measured: total cyanides and total phenol. Three-liter samples for analyses of specific organic compounds were composited in 1-gallon glass containers (no preservative added). Samples of approximately 1 liter volume were composited for T-metals (nitric acid added); T-cyanide (sodium hydroxide added); and T-phenols (phosphoric acid added).

Sample locations can be found in Figure 15. DAF effluent (sample point 1) was collected atop the aeration bays. Final clarifier effluents (points 4 and 5) were sampled from separate taps located beneath the bioreactors. Return sludge samples LAST LINE (points 2 and 3) were collected on the discharge sides of the

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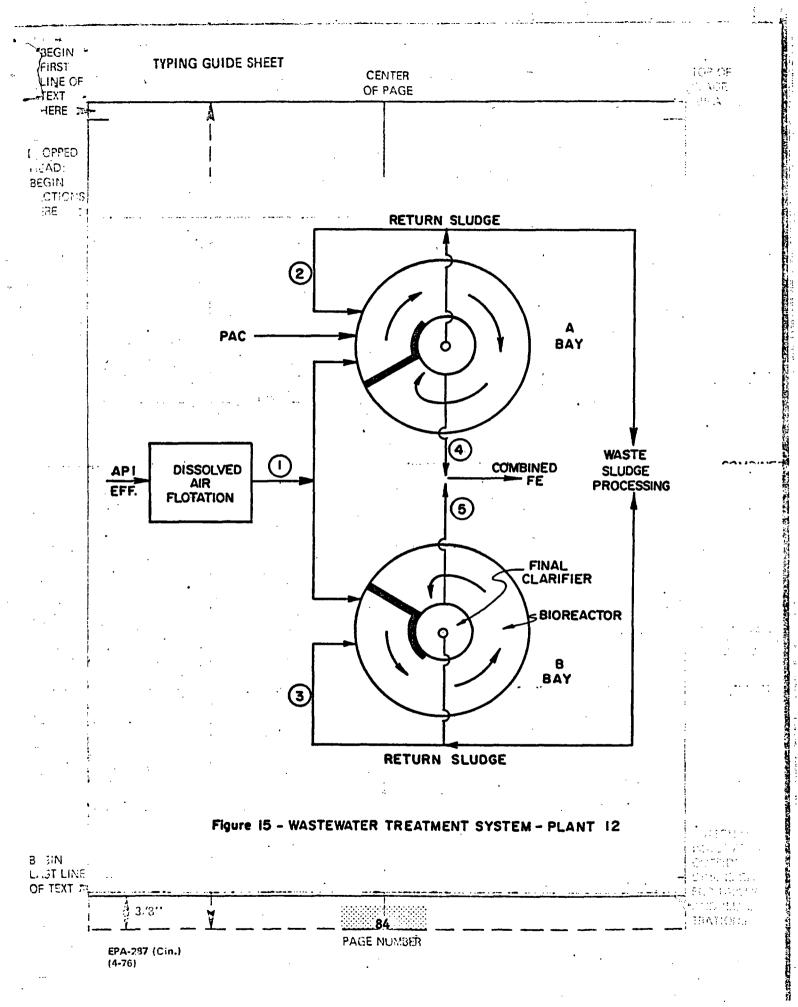
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respective lift pumps. An air-stripper sampler equipped with XAD resin trap was placed in each aeration basin at a point after entry of the DAF effluent. Volatile organics were sampled in the DAF and final clarifier effluents using the standard 28-ml VOA septum vials. The VOA grab samples were collected at 2 pm on November 8.

Samples of DAF effluent, return sludges, and final effluents were composited every 4 hours beginning at 6 pm on November 5, 1978, and ending at 2 pm on November 8, 1978. Preservatives were added at the initiation of sampling. Preservatives used were concentrated phosphoric acid to achieve a final pH \$4 (total phenol); sodium hydroxide pellets to achieve a final pH ≥12 (total cyanides); and redistilled nitric acid to achieve a final pH <2 (total metals). The compositing procedure for these samples was 55 ml every 4 hours. No preservative was used for organics sampling; a 170-ml grab sample was composited every 4 All sample containers were iced throughout the sampling period.

The air-stripper samplers were placed in operation at 9 am. November 6, and performed adequately until 2 pm, November 8, 1978, with the exception of a 4-hour stoppage on November 7 (i.e., 49 hours total operational time). Air charged to the strippers averaged 60 cfh, and the quantity of air to the XAD-2 scrubber was 30 cfh. A detailed sampling schedule is presented in Table 34. Flow data are found in Table 35.

Analytical Results--

Pollutants for which samples were collected and analyzed are presented in Table 36. All extractions and analyses of samples were conducted at RSKERL.

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TABLE 34 SAMPLING SCHEDULE (PLANT 12)

GEORGEO	Date	Time	Sample taken	Remarks
HEAD.	11/5/78	6:00 pm	Composite	Preservatives added for T-met-
DECTION:	} }	10:00 pm	11	als, T-cyanide, and T-phenols.
	11/6/78	2:00 am 6:00 am 9:00 am	11 11	Air-stripper on at 9 am.
	ATT	10:00 am 2:00 pm 6:00 pm	11 11	iii stripper on at 7 am.
		10:00 pm	11	e de la companya del companya de la companya del companya de la co
	11/7/78	2:00 am 6:00 am 10:00 am	11 17 11	Air-stripper off at 6:00 am. Air-stripper on at 10:00 am.
		- 2:00 pm 6:00 pm 10:00 pm	11	
	11/8/78	2:00 am 6:00 am 10:00 am	***	
		2:00 pm	11	Air-stripper off at 2:00 pm; VOA grabs collected at 2:00 pm

TABLE 35. DAILY FLOW DATA (PLANT 2)

Date	Return Sludge (PAC) (gpm)	Return Sludge (gpm)	Combined Final Effluent (mgd)
11/4-5/78*	500	500	2.9
11/5-6/78 11/6-7/78	500 500	500 500	4.2 3.5
11/7-3/78	500	500	3.4

\*Twenty-four hour period, from 6:00 am to 6:00 am.

Sludge wasteage was ~25 gpm, meaning that combined FE flows approximated bioinfluent (DAF) flow during the study.

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HERE B TABLE 36. ANALYTICAL DATA (PLANT 12) PAC Return Return PAC Final DROPPED Influent Sludge Sludge Effluent Effluent Priority Pollutant  $(\mu g/1)$  $(\mu g/1)$  $(\mu g/1)$ (µg/1)  $(\mu g/1)$ SECTIONS CLASSICAL TOTAL CYANIDES (mg/1)\* <.05 <.05 <.05 <.05 <.05 TOTAL PHENOL 417 386 67 42 29 TOTAL METALS Arsenic <10 260 150 <10 <10 Selenium <10 400 280 <10 <10 Cadmium <1 31 24 <1 <1 <3 Beryllium 20 9 < 3 <3 45 4,900 Copper 5,800 18 16 Antimony <10 \_<10 \_\_<10 <10 -<10 Chromium 280 64,000 60,000 100 72 Nickel 17 1,100 1,200 14 17 28,000 Zinc 390 37,000 140 130 Silver <10 22 22 <10 <10 Thallium ... <10 <10 <10 <10 <10 Lead 40 12,000 11,000 18 14 Mercury <0.6 <10 <10 <1.0 <0.8 ORGANICS (GAS CHROMATOGRAPHY) **PURGEABLES** Methylene chloride <10 N.A. <10 N.A. <10 Chloroform <10 <10 <10 Benzene 320 11 \*\* <40 <40 1,1,2,2-Tetrachloroethylene <10 11 <10 <10 17 \*\* Toluene 695 <10 <10 Ethylbenzene 56 <10 <10 POLYNUCLEAR AROMATICS Pyrene <10 · <12 <13 <10 <10 Benzo-a-pyrene <44 <106 <110 <13 <11 Chrysene <10 <14 <10 <14 <10 Fluoranthrene <10 <12 <13 <10 <10 Phenanthrene/Anthracene <10/<10 <14/<16 <14/<17 <10/<10 <10/<10 Naphthalene 282 <14 <14 <10 <10 Acenaphthene 24 <10 <10 <10 <10 Fluorene <16 <39 <41 <10 <10 SCTION OF PHENOLS 2.4-dimethylphenol 142 10 N.D. N.D. N.D. LAST LINE - Direision OF TEXT : (Continued) FORTNELL FANO RESUL TRATIONS 87

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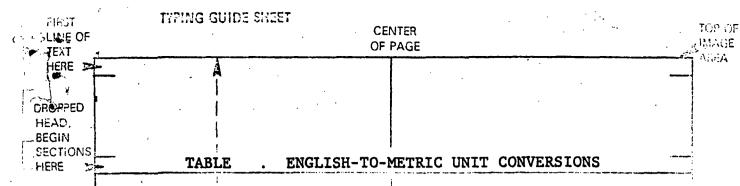
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HERE =	-		<u>.</u>	TABLE 36.	(Continued)		<del></del>		•
ROPPÉD EAD: EGIN	Pric	ority Po	ollutant	Influent (µg/1)	PAC Return Sludge (µg/1)	Return Sludge (µg/l)	PAC Effluent (µg/1)	Final Effluent (µg/l)	
ECTIONS ERE	Phthalate	. בכדדפס	i 2						• 1
	Dimo Die Di- Bis	ethyl ph thyl pht n-butyl	ithalate chalate phthalate lhexyl)						
	*Note: **Key:	N.D N.A N.S	Not Detect Not Applic No Standar	pressed in material control of the second co	s than detec	table li	mits		• · · · · · · · · · · · · · · · · · · ·
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ROPPED	Priority Pollutant	Sparged Air (µg)	, XAD-2	Sparged Air (µg)	, Tenax	
'EAD; EGIN	POLYNUCLEAR AROMATICS	PAC	<u>Control</u>			_
ECTIONS	Naphthalene	N.D.	N.D.			!
68 <b>E</b> N	2-Chloronaphthane	30	220			<del>-</del> .
7.0.1 No.	Acenaphthalene	<10	<10	····		' ·
	Acenaphthene	<10	<10			
	Fluorene	, < <b>13</b>	<13			
	Phenanthrene/Anthracene	<10	<10			
į	Fluoranthene	<10	<10			
į	Pyrene	<10	<10			
	1,2-Benzanthracene	20	<10			
	Chrysene	20	<10			
	3,4-Benzopyrene	<35	<35			
	1,2:5,6-Dibenzanthracene	N.D.	N.D.			•
	PHENOLICS		•			
		<25	<25			
i	2-Nitrophenol	<25	<25			
	Phenol	<10	<10			
į	2,4-Dimethylphenol	<25	<25			•
	2,4-Dichlorophenol	<b>&lt;50</b>	<50	•		
	2,4,6-Trichlorophenol	<10	200			
	4-Chloro-m-cresol	80	30			
	2,4-Dinitrophenol	<100	<100			
	4,6-Dinitro-o-cresol	<25	<25			
	Pentachlorophenol	<25	<25			
	4-Nitrophenol	<25	<25		•	: .
	PURGEABLES	•				-
	Methylene chloride	İ		No samples	taken.	
1	1,1-Dichloroethane	İ		-		
	1,2-Trans-dichloroethylene	:				
	Chloroform		•			
į	1,2-Dichloroethane			·		
	1,1,1-Trichloroethane					•
	Carbon tetrachloride		•			
	Dichlorobromomethane					
	1,2-Dichloropropane					:
	Benzene			•		:
_	Trichloroethylene					1
	Chlorodibromomethane					-
	1,1,2-Trichloroethane					İ
i	Methyl bromide			•		-
•	Bromoform	1				i
	1,1,2,2-Tetrachloroethane					1
l	Tetrachloroethylene					some
	Toluene					WAGE.
EGIN	Chlorobenzene			•		OUTSID
ast line <mark>i</mark>	Ethylbenzene					
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1	3.8	89		•	-	CITART



Multiply This	By This	To ObtainThis		
1bs	0.4536	kg	kilograms	
short tons	0.9072	metric tons	metric tons (1000 kg)	
short tons	907.2	kg '	kilograms	
inches	2.54	Cm.	centimetres	
feet	0.3048	m	metres	• •
statute miles	1.609	kan	kilometres	• •
gallons	3.785	1	<b>litres</b> (1000 litres = 1 m <sup>3</sup> )	
barrels	0.1590	m <sup>3</sup>	cubic metres	:
Btu	0.252	kca1	kilocalories	•
SCF	0.02679	. nm <sup>3</sup>	normal cubic metres	
Btu/1b	0.5556	kcal/kg	kilocalories/kilogram	:
Btu/CF	8.899	kca1/m <sup>3</sup>	kilocalories/cubic metre	
Btu/SCF	9.406	kca1/nm <sup>3</sup>	kilocalories/normal cubic metre	
10 <sup>9</sup> Btu/day	252	Gca1/day	gigacalories/day	
10 <sup>6</sup> Btu/day	252	Mcal/day	megacalories/day	
MM Btu/hr	252	Mca1/hr	megacalories/hour	
SCFD	0.02679	nm <sup>3</sup> /day	normal cubic metres/day	1
MM SCFD	0.02679	10 <sup>6</sup> nm <sup>3</sup> /day	million normal cubic metres/day	
		(Mnm <sup>3</sup> /day)	(mega normal cubic metres/day)	
SCF/MM Btu	0.1063	nm <sup>3</sup> /Gca1	normal cubic metres/gigacalorie	
lbs/MM Btu	1.8	kg/GCal	kilograms/gigacalorie	
1bs/CF	16.02	'kg/m <sup>3</sup>	kilograms/cubic metre	
psi .	0.07031	kg/cm <sup>2</sup> ·	kilograms/square centimetre	
gpm	0.227	m <sup>3</sup> /hr	cubic metres/hour	
acre-ft/year	0.1408	m <sup>3</sup> /hr	cubic metres/hour	
horsepower	745.7	W	watts	<u> </u>
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knot	1.852	km/hr	kilometres/hour	BOLTOM CE

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